

# Lab : Flame Test

**Main Idea :** Whoops!!! You're crazy teacher got her chemicals mixed up. It's your job to help her figure out which is which. This is what we know:

1. All of the chemicals are salts. A salt is any compound made of chloride (a halogen) and an element from groups 1 and 2 on the periodic table (Alkali & Alkaline Earth metals).
2. Some chemicals can be identified by the flame they produce. All of these compounds are **flammable**, meaning they burn. The different colors of the flames are the results of burning these different compounds.
3. The table to the right lists some chemicals and the colors they produce.

Element	Color
Barium	Yellow-green
Calcium	Orange-red
Copper	Blue-green
Lithium	Crimson Red
Potassium	Pale Violet
Sodium	Yellow
Strontium	Red

**Materials :** small beaker  
paper towels  
candle  
4 wood splints  
aluminum foil for candle wax  
safety glasses for everyone  
7 small vials of different salts

## Safety First!

1. You must wear goggles at all times during the lab – these chemicals sizzle!
2. Do NOT touch the chemicals with your skin or clothing. If you do, wash immediately.
3. Keep hair tied back and sleeves rolled up.
4. Pay attention to what you are doing and to those around you.

## Lab Procedures :

1. Read the following directions, #2 - 10, with your group before starting.
2. Fill your beaker with water and place your candle on the foil.
3. Raise your hand to begin. Your teacher will light the candle.
4. Choose one vial of "salt" from the collection. Carefully observe it with your **EYES ONLY**. Write down your observations of its physical properties on the chart.
5. **Predict the flame color** of this salt by recording your prediction on the chart.
6. Carefully have one person from your group dip the end (about 1/2 cm only) of the soaked wooden splint into the chemical so that you've collected a small amount of salt on its tip.
7. Carefully put the salt into the flame. For best results, aim the end of the stick to the side of the flame, rather than directly above it. Look very carefully, the flame will change color around the outside! It will not last long and you may only see it a little bit. **Observe and record the flame color.**
8. Dip the splint in the water. If it was not charred, you may be able to use again. Otherwise, use the other end of the splint for the next vial.
9. Repeat steps 4-8 for each vial of salt.

**Clean Up :** Throw away all used paper towels and splints. Make sure all vials are sealed tightly and placed back in the plastic bag. Place your candle and foil gently in your tub. Organize, clean counters, check bin back in to your teacher and return to your desk to work on the next few pages. Then, once settled, raise your hand to get a stamp →



# Sample High School Flame Test Lab With Answer Key

**Sebastian Brünink**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Sample High School Flame Test Lab With Answer Key:**

## Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Sample High School Flame Test Lab With Answer Key**

In a global inundated with screens and the cacophony of instant transmission, the profound energy and psychological resonance of verbal beauty frequently diminish in to obscurity, eclipsed by the regular onslaught of noise and distractions. Yet, situated within the lyrical pages of **Sample High School Flame Test Lab With Answer Key**, a interesting perform of literary beauty that pulses with fresh feelings, lies an remarkable trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, this magical opus manuals readers on a mental odyssey, lightly exposing the latent potential and profound affect stuck within the delicate web of language. Within the heart-wrenching expanse of this evocative examination, we can embark upon an introspective exploration of the book is main styles, dissect its charming publishing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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### **Table of Contents Sample High School Flame Test Lab With Answer Key**

1. Understanding the eBook Sample High School Flame Test Lab With Answer Key
  - The Rise of Digital Reading Sample High School Flame Test Lab With Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Sample High School Flame Test Lab With Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Sample High School Flame Test Lab With Answer Key
  - User-Friendly Interface

4. Exploring eBook Recommendations from Sample High School Flame Test Lab With Answer Key
  - Personalized Recommendations
  - Sample High School Flame Test Lab With Answer Key User Reviews and Ratings
  - Sample High School Flame Test Lab With Answer Key and Bestseller Lists
5. Accessing Sample High School Flame Test Lab With Answer Key Free and Paid eBooks
  - Sample High School Flame Test Lab With Answer Key Public Domain eBooks
  - Sample High School Flame Test Lab With Answer Key eBook Subscription Services
  - Sample High School Flame Test Lab With Answer Key Budget-Friendly Options
6. Navigating Sample High School Flame Test Lab With Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Sample High School Flame Test Lab With Answer Key Compatibility with Devices
  - Sample High School Flame Test Lab With Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Sample High School Flame Test Lab With Answer Key
  - Highlighting and Note-Taking Sample High School Flame Test Lab With Answer Key
  - Interactive Elements Sample High School Flame Test Lab With Answer Key
8. Staying Engaged with Sample High School Flame Test Lab With Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Sample High School Flame Test Lab With Answer Key
9. Balancing eBooks and Physical Books Sample High School Flame Test Lab With Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Sample High School Flame Test Lab With Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Sample High School Flame Test Lab With Answer Key
  - Setting Reading Goals Sample High School Flame Test Lab With Answer Key
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Sample High School Flame Test Lab With Answer Key
  - Fact-Checking eBook Content of Sample High School Flame Test Lab With Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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