

Renewing your CDA:

- Purchase a Renewal Packet

- Gather the **5** Required Requirements

- 1 Document proof of 4.5 CEUS, 3 College credits, or 45 clock hours in Early Childhood Development.

- 2 Proof of current First Aid and Infant/Child CPR certification.

- 3 Proof of individual membership to a National or Local Early Childhood membership organization.

- 4 Letter on letterhead verifying at least 80 hours of work experience with children in the past year.

- 5 Letter of Recommendation written by an Early Childhood Reviewer.

- Submit the **Renewal Application**, all 5 required renewal documents, and the \$100.00 Renewal fee to the Council.

- **Things to make the Renewal Process easier:**

- Apply on-line using YourCDA for expedited processing time.
- Purchase your packet early so that you have time to obtain/gather all required requirements.
- You can submit your completed renewal packet up to 6 months before the expiration date on your credential.
- Print clearly and remember to include an email address to receive email reminders in the future.
- The Council must received your renewal information **before** the expiration date on your credential.
- Please be aware that all renewal certificates that have an issue date of July 1, 2013 and forward are valid for three years, as of this date the Council no longer issues five year credentials.



To learn more visit the Council's website at
www.cdacouncil.org

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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of *guided inquiry designing the ideal hand warmer* - Dec 27 2022

web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$ $\Delta H_{\text{soln}} = -44.2 \text{ kJ mol rxn}$ write equations to similarly represent the dissolving process for

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state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ mol of the substance being dissolved

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