#### Class: IX

Subject: N. Science(086)

#### **Duration: 3 hours**

Maximum Marks: 80

#### General Instructions:

Read the following instructions very carefully and strictly follow them:

- (i) This question paper comprises 39 questions. All questions are compulsory.
- (ii) This question paper is divided into five sections A, B, C, D and E.
- (iii) Section A Questions No. I to 20 are multiple choice questions. Each question carries I mark.
- (iv) Section B Questions No. 21 to 26 are very short answer type questions. Each question carries 2 marks. Answer to these questions should be in the range of 30 to 50 words.
- (v) Section C Questions No. 27 to 33 are short answer type questions. Each question carries 3 marks. Answer to these questions should in the range of 50 to 80 words.
- (vi) Section D Questions No. 34 to 36 are long answer type questions. Each question carries 5 marks. Answer to these questions should be in the range of \$0 to 120 words.
- (vii) Section E Questions No. 37 to 39 are of 3 source-based/case-based units of assessment carrying 4 marks each with sub-parts.
- (viii) There is no overall choice. However, an internal choice has been provided in some sections. Only one of the alternatives has to be attempted in such questions.

			SECTION-A		
Q. No.	Questions  The similarity between solid, liquid and gas matter is that, they:  a) occupy space b) have mass c) both a and b d) have fixed shape				Marks 1
1					
2	When 20 g of potassium permanganate is dissolved in 1000ml of water, there is no increase in volume. This observation indicates that:  a) particles of water have spaces between them into which potassium permanganate particles fit.  b) particles of water attract each other.  c) particle of potassium permanganate attracts each other.  d) particles are continuously moving.				1
3	Which of the follow a) mist	ing is not a Liquid-ga b) clouds	s Aerosol: c) smoke	d) fog	1
-4	The ions present in Na <sub>2</sub> CO <sub>3</sub> are:				1

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# **Table of Contents Sample Question Paper Schools9**

- 1. Understanding the eBook Sample Question Paper Schools9
  - The Rise of Digital Reading Sample Question Paper Schools9
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Sample Question Paper Schools9
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Sample Question Paper Schools9
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Sample Question Paper Schools9
  - Personalized Recommendations
  - Sample Question Paper Schools9 User Reviews and Ratings
  - Sample Question Paper Schools9 and Bestseller Lists

- 5. Accessing Sample Question Paper Schools9 Free and Paid eBooks
  - Sample Question Paper Schools9 Public Domain eBooks
  - Sample Question Paper Schools9 eBook Subscription Services
  - Sample Question Paper Schools9 Budget-Friendly Options
- 6. Navigating Sample Question Paper Schools9 eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Sample Question Paper Schools9 Compatibility with Devices
  - Sample Question Paper Schools9 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Sample Question Paper Schools9
  - Highlighting and Note-Taking Sample Question Paper Schools9
  - Interactive Elements Sample Question Paper Schools9
- 8. Staying Engaged with Sample Question Paper Schools9
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Sample Question Paper Schools9
- 9. Balancing eBooks and Physical Books Sample Question Paper Schools9
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Sample Question Paper Schools9
- 10. Overcoming Reading Challenges
  - o Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Sample Question Paper Schools9
  - Setting Reading Goals Sample Question Paper Schools9
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sample Question Paper Schools9
  - Fact-Checking eBook Content of Sample Question Paper Schools9
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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