



Q.B. Sl. No. 373954

**POLYCET – 2014**Hall Ticket Number 

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Signature of the Candidate 

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**Time : 2 Hours****Total Marks : 120****Note :** Before answering the questions, read carefully the instructions given on the OMR sheet

ప్రశ్నలకు జవాబులు ప్రాయోగికము ముందు OMR జవాబు పత్రములో జన్యపరిచిన సూచనలను జాగ్రత్తగా చదవండి.

**SECTION - I : MATHEMATICS**

1. In a triangle  $ABC$ ,  $\angle A = 90^\circ$ ,  $\overline{AD} \perp \overline{BC}$ , if  $AC = 10\sqrt{3}$ ,  $BD = 5$ ,  $DC = x$ , then the value of  $x$  is  
 $ABC$  త్రిభుజములో  $\angle A = 90^\circ$ ,  $\overline{AD} \perp \overline{BC}$ ,  $AC = 10\sqrt{3}$ ,  $BD = 5$ ,  $DC = x$  అయితే  $x$  విలువ =
- (1) 15                      (2) 10                      (3) 20                      (4)  $5\sqrt{3}$

2. If  $(1, 2)$ ,  $(-1, 4)$  and  $(7, -k)$  are collinear, then  $k =$   
 $(1, 2)$ ,  $(-1, 4)$  మరియు  $(7, -k)$  లు సరేఖీయాలయితే  $k$  విలువ =
- (1) 1                      (2) -1                      (3) 2                      (4) 4

3. The slope of the line which makes  $\frac{3\pi}{4}$  with the +ve direction of  $x$ -axis is

$x$  - అక్షముతో  $\frac{3\pi}{4}$  చేస్తున్న రేఖలో  $\frac{3\pi}{4}$  కోణము చేసే సరేఖలో వాలు =

- (1) 1                      (2) -1                      (3) 2                      (4) 3

4. The nature of the lines  $7x + 5y + 6 = 0$  and  $5x - 7y + 6 = 0$  is  
 $7x + 5y + 6 = 0$  మరియు  $5x - 7y + 6 = 0$  రేఖలు

- (1) parallel.                      (2) concurrent  
 సమాంతరాలు                      ఋణిత రేఖలు  
 (3) perpendicular                      (4) None  
 లంబ రేఖలు                      ఏదీ కాదు

SPACE FOR ROUGH WORK / చిత్తపనికి కేటాయించబడిన స్థలము

# Polytechnic Entrance Exam Question Paper

**Hussin A.Rothana**



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of all individual orders the rate of a reaction is

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web oct 27 2022 the rate law for the reaction ce h<sub>2</sub> g ce 2no g ce n<sub>2</sub>o g ce h<sub>2</sub>o g nonumber has been experimentally determined to be rate

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web 1 use table 1 to determine the reaction order for hcl 2 use table 2 to determine the reaction order for  $\text{Na}_2\text{S}_2\text{O}_3$  remember you want to see what happens to the reaction

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