

## Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

OH  
NO<sub>3</sub>  
C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

hydroxide  
nitrate  
acetate

ClO<sub>3</sub>  
HCO<sub>3</sub>  
SO<sub>4</sub>  
CO<sub>3</sub>

chlorate  
bicarbonate  
sulfate  
carbonate

SO<sub>3</sub>  
PO<sub>4</sub>  
NH<sub>4</sub>

sulfite  
phosphate  
ammonium

- |  |   |
|--|---|
| 1. CuSO <sub>4</sub> _____                             | 26. NaOH _____  |
| 2. KClO <sub>3</sub> _____                             | 27. NaNO <sub>3</sub> _____                               |
| 3. NH <sub>4</sub> OH _____                            | 28. Ba(OH) <sub>2</sub> _____                             |
| 4. K <sub>2</sub> CO <sub>3</sub> _____                | 29. K <sub>2</sub> CO <sub>3</sub> _____                  |
| 5. Na <sub>2</sub> SO <sub>4</sub> _____               | 30. LiOH _____  |
| 6. KC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> _____ | 31. KHCO <sub>3</sub> _____                               |
| 7. NH <sub>4</sub> Cl _____                            | 32. NH <sub>4</sub> F _____                               |
| 8. Mg(OH) <sub>2</sub> _____                           | 33. KClO <sub>3</sub> _____                               |
| 9. Zn(NO <sub>3</sub> ) <sub>2</sub> _____             | 34. NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> _____  |
| 10. MgSO <sub>4</sub> _____                            | 35. CaClO <sub>3</sub> _____                              |
| 11. CaCO <sub>3</sub> _____                            | 36. SrNO <sub>3</sub> _____                               |
| 12. KOH _____  | 37. Be(OH) <sub>2</sub> _____                             |
| 13. Ca(OH) <sub>2</sub> _____                          | 38. Li <sub>2</sub> SO <sub>4</sub> _____                 |
| 14. K <sub>2</sub> CO <sub>3</sub> _____               | 39. Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> _____ |
| 15. AlPO <sub>4</sub> _____                            | 40. BaSO <sub>3</sub> _____                               |
| 16. Na <sub>2</sub> PO <sub>4</sub> _____              | 41. NH <sub>4</sub> I _____                               |
| 17. MgSO <sub>4</sub> _____                            | 42. NH <sub>4</sub> HCO <sub>3</sub> _____                |
| 18. AgNO <sub>3</sub> _____                            | 43. Cu <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub> _____ |
| 19. Na <sub>2</sub> CO <sub>3</sub> _____              | 44. Ni(OH) <sub>2</sub> _____                             |
| 20. NaHCO <sub>3</sub> _____                           | 45. ZnSO <sub>4</sub> _____                               |
| 21. Ca(NO <sub>3</sub> ) <sub>2</sub> _____            | 46. MnCO <sub>3</sub> _____                               |
| 22. BaSO <sub>4</sub> _____                            | 47. H <sub>2</sub> SO <sub>4</sub> _____                  |
| 23. BPO <sub>3</sub> _____                             | 48. Au <sub>2</sub> SO <sub>4</sub> _____                 |
| 24. LiNO <sub>3</sub> _____                            | 49. NH <sub>4</sub> NO <sub>3</sub> _____                 |
| 25. Al(NO <sub>3</sub> ) <sub>3</sub> _____            | 50. LiClO <sub>3</sub> _____                              |

# Polyatomic Ions Naming Answer

**Katrin Zwirgmaier**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Polyatomic Ions Naming Answer:**

Chemistry: 1001 Practice Problems For Dummies (+ Free Online Practice) Heather Hattori, Richard H.

Langley, 2022-06-08 Practice your way to a better grade in your Chemistry class Chemistry 1001 Practice Problems For Dummies gives you 1 001 opportunities to practice solving problems on all the topics covered in your chemistry class in the book and online Get extra practice with tricky subjects solidify what you've already learned and get in depth walk throughs for every problem with this useful book These practice problems and detailed answer explanations will catalyze the reactions in your brain no matter what your skill level Thanks to Dummies you have a resource to help you put key concepts into practice Work through multiple choice practice problems on all Chemistry topics covered in class Step through detailed solutions to build your understanding Access practice questions online to study anywhere any time Improve your grade and up your study game with practice practice practice The material presented in Chemistry 1001 Practice Problems For Dummies is an excellent resource for students as well as parents and tutors looking to help supplement classroom instruction Chemistry 1001 Practice Problems For Dummies 9781119883531 was previously published as 1 001 Chemistry Practice Problems For Dummies 9781118549322 While this version features a new Dummies cover and design the content is the same as the prior release and should not be considered a new or updated product

### **Chemistry: Concepts and Problems**

Clifford C. Houk, Richard Post, 1996-03-09 CHEMISTRY SECOND EDITION The fast easy way to master the fundamentals of chemistry Have you ever wondered about the differences between liquids gases and solids Or what actually happens when something burns What exactly is a solution An acid A base This is chemistry the composition and structure of substances composing all matter and how they can be transformed Whether you are studying chemistry for the first time on your own want to refresh your memory for a test or need a little help for a course this concise interactive guide gives you a fresh approach to this fascinating subject This fully up to date edition of Chemistry Concepts and Problems Has been tested rewritten and retested to ensure that you can teach yourself all about chemistry Requires no prerequisites Lets you work at your own pace with a helpful question and answer format Lists objectives for each chapter you can skip ahead or find extra help if you need it Reinforces what you learn with chapter self tests

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*Study Guide to Accompany Basics for Chemistry* Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular

examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

**Mapping College Chemistry** Stephen DeMeo, 2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry Topical problem solving will focus on limiting reactant stoichiometry identifying types of chemical reactions equilibrium acid base equilibria and electrochemistry Why would this resource be of interest to chemistry students To be successful to get into a well known college medical school physical therapy or graduate program often requires that students get an A in your pre requisite Introductory General Chemistry course To make matters worse many college professors feel that only a few students should get A grades and therefore they give difficult exams that many students fail this is the weeding out process that every pre health student is apprehensive about To succeed in this competitive environment entails not just studying harder or longer it means re organizing textbook content so that it is meaningful to the student This is the first text of its kind to employ a reliable research based strategy that incorporates a decision based visual tool to solve chemistry textbook problems ones that can make or break a career

Foundations of College Chemistry, Alternate Morris Hein, Susan Arena, 2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

*Survival Guide to General Chemistry* Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available

specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields **General, Organic, and Biological Chemistry** Michael P. Garoutte,2014-02-24

Classroom activities to support a General Organic and Biological Chemistry text Students can follow a guided inquiry approach as they learn chemistry in the classroom General Organic and Biological Chemistry A Guided Inquiry serves as an accompaniment to a GOB Chemistry text It can suit the one or two semester course This supplemental text supports Process Oriented Guided Inquiry Learning POGIL which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration *Introductory Chemistry* Michael P. Garoutte,Ashley B.

Mahoney,2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting **Let's Review**

**Regents: Chemistry--Physical Setting Revised Edition** Barron's Educational Series,Albert S. Tarendash,2021-01-05 Barron's Let's Review Regents Chemistry gives students the step by step review and practice they need to prepare for the Regents Chemistry Physical Setting exam This updated edition is an ideal companion to high school textbooks and covers all Chemistry topics prescribed by the New York State Board of Regents Let's Review Regents Chemistry covers all high school level Chemistry topics and includes Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key *CliffsNotes Chemistry Practice Pack* Charles Henrickson,2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations

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