

Name \_\_\_\_\_ answers \_\_\_\_\_ Date \_\_\_\_\_

# FORMULAS WITH POLYTOMIC IONS

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions.

	$\text{OH}^-$	$\text{OH}_3^-$	$\text{CO}_3^{+}$	$\text{SO}_4^{+}$	$\text{PO}_4^{+}$
$\text{H}^+$	$\text{HOH}$ $(\text{H}_2\text{O})$	$\text{HNO}_3$	$\text{H}_2\text{CO}_3$	$\text{H}_2\text{SO}_4$	$\text{H}_3\text{PO}_4$
$\text{Na}^+$	$\text{NaOH}$	$\text{NaNO}_3$	$\text{Na}_2\text{CO}_3$	$\text{Na}_2\text{SO}_4$	$\text{Na}_3\text{PO}_4$
$\text{Mg}^{+2}$	$\text{Mg}(\text{OH})_2$	$\text{Mg}(\text{NO}_3)_2$	$\text{MgCO}_3$	$\text{MgSO}_4$	$\text{Mg}_3(\text{PO}_4)_2$
$\text{NH}_4^{+}$	$\text{NH}_4\text{OH}$	$\text{NH}_4\text{NO}_3$	$(\text{NH}_4)_2\text{CO}_3$	$(\text{NH}_4)_2\text{SO}_4$	$(\text{NH}_4)_3\text{PO}_4$
$\text{Ca}^{+2}$	$\text{KOH}$	$\text{KNO}_3$	$\text{K}_2\text{CO}_3$	$\text{K}_2\text{SO}_4$	$\text{K}_3\text{PO}_4$
$\text{K}^+$	$\text{Al}(\text{OH})_3$	$\text{Al}(\text{NO}_3)_3$	$\text{Al}_2(\text{CO}_3)_3$	$\text{Al}_2(\text{SO}_4)_3$	$\text{AlPO}_4$
$\text{Al}^{+3}$	$\text{Pb}(\text{OH})_4$	$\text{Pb}(\text{NO}_3)_4$	$\text{Pb}(\text{CO}_3)_2$	$\text{Pb}(\text{SO}_4)_2$	$\text{Pb}_3(\text{PO}_4)_4$

# Polyatomic Ionic Formulas Answers

**Rosina Ehmman**



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## **Table of Contents Polyatomic Ionic Formulas Answers**

1. Understanding the eBook Polyatomic Ionic Formulas Answers
  - The Rise of Digital Reading Polyatomic Ionic Formulas Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ionic Formulas Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in a Polyatomic Ionic Formulas Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ionic Formulas Answers
  - Personalized Recommendations
  - Polyatomic Ionic Formulas Answers User Reviews and Ratings
  - Polyatomic Ionic Formulas Answers and Bestseller Lists
5. Accessing Polyatomic Ionic Formulas Answers Free and Paid eBooks
  - Polyatomic Ionic Formulas Answers Public Domain eBooks
  - Polyatomic Ionic Formulas Answers eBook Subscription Services
  - Polyatomic Ionic Formulas Answers Budget-Friendly Options

6. Navigating Polyatomic Ionic Formulas Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Polyatomic Ionic Formulas Answers Compatibility with Devices
  - Polyatomic Ionic Formulas Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Polyatomic Ionic Formulas Answers
  - Highlighting and Note-Taking Polyatomic Ionic Formulas Answers
  - Interactive Elements Polyatomic Ionic Formulas Answers
8. Staying Engaged with Polyatomic Ionic Formulas Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Polyatomic Ionic Formulas Answers
9. Balancing eBooks and Physical Books Polyatomic Ionic Formulas Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Polyatomic Ionic Formulas Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Polyatomic Ionic Formulas Answers
  - Setting Reading Goals Polyatomic Ionic Formulas Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Polyatomic Ionic Formulas Answers
  - Fact-Checking eBook Content of Polyatomic Ionic Formulas Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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