

Polyatomic Ion Worksheet

Place the charges on each ion – Write the Formula – Name the Compound

Remember – The cation always goes first.

Example:	NH_4^{+1}	&	I^{-1}	Formula:	<u>NH_4I</u>	Name:	<u>Ammonium Iodide</u>
1.	NH_4	&	O	Formula:	_____	Name:	_____
2.	Ca	&	CO_3	Formula:	_____	Name:	_____
3.	HCO_3	&	Na	Formula:	_____	Name:	_____
4.	NO_3	&	Mg	Formula:	_____	Name:	_____
5.	K	&	SO_4	Formula:	_____	Name:	_____
6.	Li	&	PO_4	Formula:	_____	Name:	_____

Finding the Formula from the Name

Example:	Lithium Chloride	Ions:	<u>Li^{+1} & Cl^{-1}</u>	Formula:	<u>LiCl</u>
1.	Sodium Oxide	Ions:	_____	Formula:	_____
2.	Potassium Bromide	Ions:	_____	Formula:	_____
3.	Calcium Nitride	Ions:	_____	Formula:	_____
4.	Beryllium Sulfide	Ions:	_____	Formula:	_____
5.	Magnesium Iodide	Ions:	_____	Formula:	_____
6.	Aluminum Oxide	Ions:	_____	Formula:	_____
7.	Ammonium Bromide	Ions:	_____	Formula:	_____
8.	Calcium Bicarbonate	Ions:	_____	Formula:	_____
9.	Potassium Phosphate	Ions:	_____	Formula:	_____
10.	Aluminum Sulfate	Ions:	_____	Formula:	_____

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K. Kuchitsu



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web mar 12 2021 it was here that lewis s career in developmental biology took off his engineering and mathematical training allowed him to be active in modelling various developmental events and was part of an influential group of theoretical biologists that included c h waddington