

PHYSICAL SCIENCES P.2

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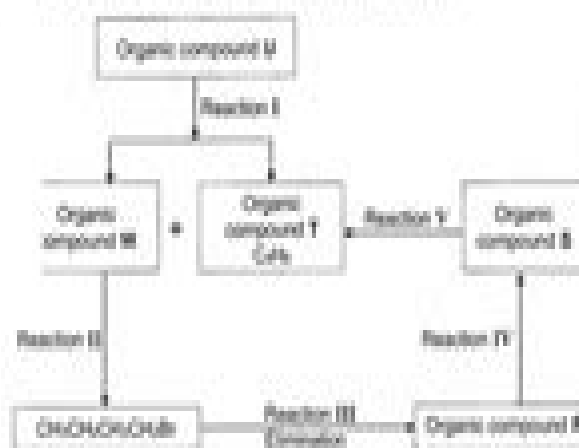
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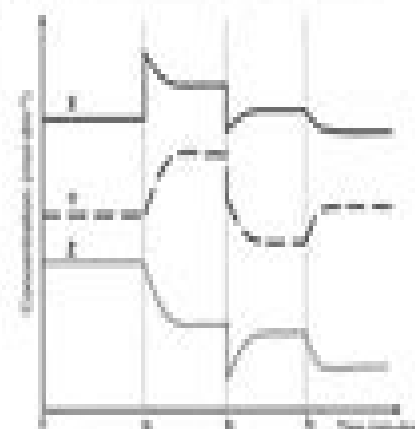
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Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

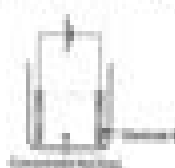
The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reagent used in reaction IV. (1 mark)



4.8 Draw the half-cell reaction for the zinc electrode. (1 mark)

4.9 Write the overall cell reaction. (1 mark)

4.10 Explain the reason for the following observations. (2 marks)



4.11 Explain the reason for the following observations. (2 marks)

4.12 The diagram shows a beaker containing a solution and a piece of metal (labeled 'Metal') that is partially submerged. Bubbles are shown forming on the surface of the metal.

4.13 The diagram shows a beaker containing a solution and a piece of metal (labeled 'Metal') that is partially submerged. Bubbles are shown forming on the surface of the metal.

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Siddappa N.Byrareddy



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