

Periodic Table Basics

Step 1: Complete the square for each element by filling in the atomic number, name, & atomic mass.

Step 2: Determine the number of protons, neutrons, and electrons in an atom of each element.

Step 3: Identify if the element is a solid, liquid, or gas at room temperature.

Step 4: Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

Step 5: List at least three physical or chemical properties for each element.

Step 6: List at least three uses for each element.

Step 7: Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

Step 8: Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na
Orange = B & Al

Pink = O & S
Red = C & Si

Blue = Be & Mg
Tan = N & P

Purple = F & Cl
Yellow = He, Ne, & Ar

Step 9: Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

① <div style="text-align: center; font-size: 2em;">B</div> 	P= ② N= _____ E= _____	③ O S O L O G
	M.P. = _____ B.P. = ④ _____	
Properties ⑤ Uses ⑥ Bohr Diagram Lewis Structure <div style="display: flex; align-items: center; justify-content: space-around;">  <div style="text-align: center;"> ⑦ B </div> </div>		

Physical Science If8767 Page 64 Answer Worksheet

**David Frank, Michael Wyssession, Sophia
Yancopoulor**

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Table of Contents Physical Science If8767 Page 64 Answer Worksheet

1. Understanding the eBook Physical Science If8767 Page 64 Answer Worksheet
 - The Rise of Digital Reading Physical Science If8767 Page 64 Answer Worksheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Page 64 Answer Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science If8767 Page 64 Answer Worksheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Page 64 Answer Worksheet

- Personalized Recommendations
 - Physical Science If8767 Page 64 Answer Worksheet User Reviews and Ratings
 - Physical Science If8767 Page 64 Answer Worksheet and Bestseller Lists
5. Accessing Physical Science If8767 Page 64 Answer Worksheet Free and Paid eBooks
 - Physical Science If8767 Page 64 Answer Worksheet Public Domain eBooks
 - Physical Science If8767 Page 64 Answer Worksheet eBook Subscription Services
 - Physical Science If8767 Page 64 Answer Worksheet Budget-Friendly Options
 6. Navigating Physical Science If8767 Page 64 Answer Worksheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science If8767 Page 64 Answer Worksheet Compatibility with Devices
 - Physical Science If8767 Page 64 Answer Worksheet Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science If8767 Page 64 Answer Worksheet
 - Highlighting and Note-Taking Physical Science If8767 Page 64 Answer Worksheet
 - Interactive Elements Physical Science If8767 Page 64 Answer Worksheet
 8. Staying Engaged with Physical Science If8767 Page 64 Answer Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science If8767 Page 64 Answer Worksheet
 9. Balancing eBooks and Physical Books Physical Science If8767 Page 64 Answer Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science If8767 Page 64 Answer Worksheet
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science If8767 Page 64 Answer Worksheet
 - Setting Reading Goals Physical Science If8767 Page 64 Answer Worksheet
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science If8767 Page 64 Answer Worksheet

- Fact-Checking eBook Content of Physical Science If8767 Page 64 Answer Worksheet
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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