# WORD EQUATIONS Name Write and balance the following chemical equations. Hydrogen plus oxygen vield water. Nitrogen plus hydrogen yield ammonia. Aluminum bromide plus chlorine yield aluminum chloride and bromine. Hydrochloric acid plus sodium hydroxide yield sodium chloride plus water. Iron plus lead (II) sulfate react forming iron (II) sulfate plus lead. Potassium chlorate when heated produces potassium chloride plus oxygen gas. Sulfuric acid decomposes to form sulfur trioxide gas plus water. Sodium oxide combines with water to make sodium hydroxide.

- Potassium lodide reacts with bromine forming potassium bromide plus lodine.
- Sodium phosphate reacts with calcium nitrate to produce sodium nitrate plus 10. calcium phosphate.
- 11. Zinc reacts with iron (III) chloride yielding zinc chloride plus iron precipitate.
- 12. Ammonium carbonate and magnesium sulfate react to yield ammonium sulfate plus magnesium carbonate.
- Phosphoric acid plus calcium hydroxide react forming solid calcium phosphate plus 13. water.
- 14. Aluminum plus oxygen gas form aluminum oxide under certain conditions.
- 15. Nitrogen gas plus oxygen gas react and form dinitrogen pentoxide.

# **Physical Science If8767 Word Equations**

**Thomas Griffiths** 

**Physical Science If8767 Word Equations:** 

## Physical Science If8767 Word Equations Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Physical Science If8767 Word Equations**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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