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The data presented in tables 1 and 2 are the means of the measurements. The mean body temperatures were not significantly different between the control and treatment groups. The mean body temperatures were 37.0°C and 37.1°C, respectively.

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Source: National Health and Medical Research Council (NH&MRC) (2000). *Guidelines for the management of acute otitis media with effusion*. Canberra: Australian Government Publishing Service. Available at: <http://www.nhmrc.gov.au/publications/otitismedia.htm>

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For a full discussion of the legal and ethical issues surrounding the use of genetic testing, see the article by Dr. David Asch, "Genetic Testing: A New Frontier in Medicine," *Journal of the American Medical Association*, 2000; 284:1000-1005.

1. **Chlorine** is the most common disinfectant used in water treatment. It is added to water in the form of chlorine gas, sodium hypochlorite (bleach), or calcium hypochlorite (bleaching powder). Chlorine kills bacteria, viruses, and protozoa by oxidizing their cell walls and internal structures. The typical dosage is 1-2 mg/L.
 2. **Ozone** is a powerful oxidizing agent that kills microorganisms by attacking their cell membranes and DNA. It is used in some advanced water treatment systems. The typical dosage is 1-3 mg/L.
 3. **Ultraviolet (UV) light** is used to kill microorganisms by damaging their DNA. It is effective against bacteria, viruses, and protozoa. The typical dosage is 40,000 mJ/m².
 4. **Chloramines** are formed by the reaction of chlorine and ammonia. They are used as a secondary disinfectant to provide a longer-lasting residual in the distribution system. The typical dosage is 1-2 mg/L.
- Disinfection by-products (DBPs)** are formed when disinfectants react with natural organic matter (NOM) in water. Some DBPs are harmful to human health, so it is important to monitor and control their formation.
1. **Trihalomethanes (THMs)** are the most common DBPs. They are formed when chlorine reacts with NOM. High levels of THMs can cause cancer and other health problems.
 2. **Halooxymethanes (HOMs)** are also common DBPs. They are formed when chlorine reacts with NOM. HOMs can cause cancer and other health problems.
 3. **Haloketones (HKs)** are another class of DBPs. They are formed when chlorine reacts with NOM. HKs can cause cancer and other health problems.
 4. **Halonitriles (HNs)** are also a class of DBPs. They are formed when chlorine reacts with NOM. HNs can cause cancer and other health problems.



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1. **Explain the difference between a primary and a secondary source of information.**
2. **Identify the primary and secondary sources of information in the following list.**
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4. **Identify the primary and secondary sources of information in the following list.**
5. **Explain the difference between a primary and a secondary source of information.**
6. **Identify the primary and secondary sources of information in the following list.**
7. **Explain the difference between a primary and a secondary source of information.**
8. **Identify the primary and secondary sources of information in the following list.**
9. **Explain the difference between a primary and a secondary source of information.**
10. **Identify the primary and secondary sources of information in the following list.**

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72. *Redding-Williams* and *Williams* propose a framework around four key issues for the development of a national system for accreditation of nursing education. *See* [Redding-Williams and Williams, 2003](#).

4. [Scatterplot with fitted regression line](#) in [ggplot2](#) (https://www.datacamp.com/courses/data-manipulation-with-ggplot2/lesson-10)
5. [Boxplot](#) in [ggplot2](#) (https://www.datacamp.com/courses/data-manipulation-with-ggplot2/lesson-10)
6. [ggplot2: faceting](#) (https://www.datacamp.com/courses/data-manipulation-with-ggplot2/lesson-10)
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Nokia 6085 User Guide

Padhraic Smyth



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