

Polyatomic Ions WS

Ammonium (NH_4)¹⁺
 Carbonate (CO_3)²⁻
 Chromate (CrO_4)²⁻

Dichromate (Cr_2O_7)²⁻
 Hydroxide (OH)¹⁻
 Nitrate (NO_3)¹⁻

Nitrite (NO_2)¹⁻
 Sulfate (SO_4)²⁻
 Sulfite (SO_3)²⁻

Phosphate (PO_4)³⁻

Write the NAME of each of the following compounds.

1. $(\text{NH}_4)\text{Cl}$ Ammonium Chloride
2. $\text{Be}(\text{SO}_4)$ Beryllium Sulfate
3. $(\text{NH}_4)_3\text{N}$ Ammonium Nitride
4. MgCl_2 Magnesium Chloride
5. $\text{NH}_4(\text{NO}_3)$ Ammonium Nitrate
6. $\text{Sr}_3(\text{PO}_4)_2$ Strontium Phosphate
7. $\text{Zn}(\text{CrO}_4)_2$ Zinc Dichromate
8. $\text{K}_2(\text{Cr}_2\text{O}_7)$ Potassium Dichromate

9. $\text{Ga}(\text{ClO}_3)_3$ Gallium Chlorate
10. $\text{Cu}(\text{OH})$ Copper I Hydroxide
11. $(\text{NH}_4)_3(\text{PO}_4)$ Ammonium Phosphate
12. $\text{Fe}(\text{SO}_4)$ Iron II Sulfate
13. $\text{Mg}(\text{NO}_2)_2$ Magnesium Nitrite
14. $(\text{NH}_4)\text{NO}_2$ Ammonium Nitrite
15. $\text{Na}_2(\text{Cr}_2\text{O}_7)$ Sodium Dichromate
16. $\text{Na}(\text{OH})$ Sodium Hydroxide

Write the CHEMICAL FORMULA for each of the given NAMES.....cross charges

- | | | | | | |
|-------------------------|---------------------------------------|---|-------------------------------|---|-----------------|
| 17. calcium carbonate | Ca^{2+} $(\text{CO}_3)^{2-}$ | = | $\text{Ca}_2(\text{CO}_3)_2$ | = | CaCO_3 |
| 18. barium nitrate | Ba^{+2} NO_3^{-1} | = | $\text{Ba}(\text{NO}_3)$ | = | |
| 19. ammonium sulfate | NH_4^{+1} SO_4^{-2} | = | $(\text{NH}_4)_2 \text{SO}_4$ | = | |
| 20. aluminum hydroxide | Al^{+3} OH^{-1} | = | $\text{Al}(\text{OH})_3$ | = | |
| 21. calcium phosphate | Ca^{+2} PO_4^{-3} | = | $\text{Ca}_3(\text{PO}_4)_2$ | = | |
| 22. cesium nitrate | Cs NO_2^{-1} | = | CsNO_3 | = | |
| 23. sodium nitrite | Na^{+1} NO_3^{-1} | = | NaNO_3 | = | |
| 24. calcium sulfate | Ca^{+2} SO_4^{-2} | = | CaSO_4 | = | |
| 25. beryllium sulfate | Be SO_4^{-2} | = | BeSO_4 | = | |
| 26. sodium carbonate | Na^{+1} CO_3^{-2} | = | Na_2CO_3 | = | |
| 27. magnesium phosphate | Mg^{+2} PO_4^{-3} | = | $\text{Mg}_3(\text{PO}_4)_2$ | = | |
| 28. calcium phosphate | Ca^{+2} PO_4^{-3} | = | $\text{Ca}_3(\text{PO}_4)_2$ | = | |

Polyatomic Ions Ws

Patrick Vollmar

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Polyatomic Ions Ws:

Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **Polyatomic Ions Ws**

In some sort of inundated with displays and the cacophony of quick conversation, the profound energy and mental resonance of verbal beauty usually fade into obscurity, eclipsed by the constant onslaught of noise and distractions. However, situated within the musical pages of **Polyatomic Ions Ws**, a interesting function of literary beauty that impulses with fresh thoughts, lies an wonderful journey waiting to be embarked upon. Published by way of a virtuoso wordsmith, that interesting opus instructions visitors on a psychological odyssey, lightly revealing the latent potential and profound impact stuck within the complex web of language. Within the heart-wrenching expanse of the evocative analysis, we can embark upon an introspective exploration of the book is main subjects, dissect their charming writing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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web spirits of the adults at midnight of the following day november 2nd the celebrations shift to honor the lives of the departed adults the night is filled with laughter and fun memories much like the night before however the ofrendas take on a more adult like theme with tequila pan de muerto mezcal pulque and jars of atole

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