

Chapter 6 Chemical Bonds

# Word Wise

Use the terms from the following list to fit each of the clues given below.

molecular	covalent	polyatomic
crystals	anion	polar
ionic	alloy	metallic
formula	bond	cation

## Clues

## Vocabulary Terms

A type of bond that holds cations and anions together

A type of bond in which two atoms share a pair of valence electrons

A neutral group of atoms that are joined together by one or more covalent bonds

A term describing a covalent bond in which electrons are not shared equally

An ion that contains a covalently bonded group of atoms

An ion with a negative charge

An ion with a positive charge

A notation that shows what elements a compound contains and the ratio of the atoms or ions of these elements in the compound

Solids whose particles are arranged in a lattice structure

A mixture of two or more elements, at least one of which is a metal

A type of bond that exists between a metal cation and the shared electrons that surround it

The force that holds atoms or ions together

If you finish, try solving a few extra at the end.

# Physical Science Chapter 6 Chemical Bonds Wordwise Answers

**Marco Cascella**



**Physical Science Chapter 6 Chemical Bonds Wordwise Answers:**

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## Reviewing **Physical Science Chapter 6 Chemical Bonds Wordwise Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Physical Science Chapter 6 Chemical Bonds Wordwise Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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