

Physical Science Chapter 11 Test Review

- Two children fight over a toy, pulling in opposite directions. The first pulls East with a force of 20N. The second child pulls West with a force of 40N. What is the net force on the toy?
 - 20 N West
- A net force will cause an object to
 - Accelerate (speed up, slow down, change direction)
- A Newton is a unit of _____ and is equal to 1 _____.
 - Force; kg-m/s
- Newton's First Law is also called the Law of _____.
 - Inertia
- In order to find the force of an object, you must know its
 - Mass and acceleration
- According to Newton's First Law, a book on a shelf will stay put unless
 - A net force or unbalanced force acts on it
- According to Newton's First Law, a ball rolling down the street would roll forever. In reality, the ball stops because
 - A net force (friction) acts on the ball
- According to Newton's Second Law, force can be increased by
 - Increasing mass or increasing acceleration
- A kid stands on a skateboard and throws a baseball to a friend. According to Newton's Third Law, the skateboard will
 - Move backwards away from the throw with a force equal to the throw
- A pencil is dropped on the floor. According to Newton's Third Law, the floor and pencil
 - Hit each other with equal and opposite forces
- A Ninja karate chops a brick. What is the action force?
 - Ninja's hand on the brick
- You and a friend decide to go to Target and have a shopping cart race. When planning your strategy, you decide to choose an empty cart because
 - Less mass means less inertia, which makes the cart easier to accelerate (speed up, slow down, change direction)
- Anything with mass has
 - gravity
- As you get closer to Earth's surface, the force of _____ becomes stronger.
 - gravity
- In a plane, you weigh less because
 - You are farther from earth's surface and gravity is less
- A soccer player kicks a ball. The reaction force is
 - Ball's force on the kicker's foot
- By the year 2050, families will vacation on the moon. While on vacation with your children, you will
 - Have the same mass but weigh less
- A _____ is a unit of weight and a _____ is a unit of mass.
 - Newton, kilogram (or gram, milligram)
- An elephant and his friend the mouse go sky diving together on the moon. Since there is no air resistance, the
 - Elephant and mouse hit the surface together

Physical Science Chapter 11 Test

RC Schank



Physical Science Chapter 11 Test:

Unveiling the Magic of Words: A Overview of "**Physical Science Chapter 11 Test**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Physical Science Chapter 11 Test**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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 $\text{GaBr} + \text{NaBr} \rightarrow \text{SnO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{S}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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 $3\text{HBr} + 1\text{KHCO}_3 \rightarrow 1\text{H}_2\text{O} + 1\text{KBr} + \text{CO}_2$
 $2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{NaBr} + 3\text{SNO}_2 + 3\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$ using the following equation
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu} + \text{Zn} + \text{CuCl}_2$
 how many moles of $\text{Ce} + \text{ZnCl}_2 + \text{ZnCl}_2$ will be produced from 23.0 g of $\text{Ce} + \text{Zn}$ assuming $\text{Ce} + \text{CuCl}_2 + \text{CuCl}_2$ is available in excess

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 $\text{Br}_2 + \text{N}_2 + 3\text{F}_2 + 2\text{NF}_3 \rightarrow \text{Br}_2 + 2\text{C}_6\text{H}_{10} + 17\text{O}_2 + 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $\text{Br}_2 + 3\text{HBr} + 1\text{KHCO}_3 + 1\text{H}_2\text{O} + 1\text{KBr} + \text{CO}_2 \rightarrow \text{Br}_2 + 4\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{SO}_3 + 3\text{NaBr} + 5\text{SNO}_2 + 3\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$ using the following equation
 $\text{Br}_2 + 2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4 + \text{Br}_2$

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 $\text{SiC} \rightarrow \text{SiC}$
 answer $\text{Ce} + \text{SiO}_2 + 3\text{C} \rightarrow \text{SiC} + 2\text{CO}$

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