

OUTLINE - SENIOR HONOR'S THESIS

PURPOSE: To show that there is a moral obligation on the part of society as a whole to alleviate the suffering of the citizens living in the society.

I. INTRODUCTION

- A. Illustrative scene in which a person is depicted in a state of suffering.
- B. Pose the question as to whether or not the society this person lives in is obligated in any way to see to the welfare of this person.
- C. Brief overview of the paper.

II. THE NATURE AND MORAL WEIGHT OF SUFFERING

- A. The Nature of Suffering
 - 1. Mayerfeld's definition - suffering as a "bad overall feeling."
 - 2. Objection to Mayerfeld
 - 3. Description of suffering as the blocking off of one or more of Maslow's hierarchical needs
- B. Why does suffering carry moral weight?
 - 1. Traditional view of suffering as being inherently bad (Utilitarian account shared by Mayerfeld.)
 - 2. The Kantian view - that by allowing person x to experience suffering, that person is being denied the possibility to pursue higher ends.
- C. Transition: Now that we have covered the nature of suffering and the moral weight involved with suffering, we will take a look at the moral theories involved.

III. RELATED MORAL THEORIES IN THE CONCEPT OF ALLEVIATING SUFFERING

- A. Utilitarianism
 - 1. Peter Singer's argument from "Famine, Affluence, and Morality."
 - 2. Mayerfeld's argument from Suffering and Moral Responsibility
 - 3. Where these two fail in providing more than a subjective individual argument to assist in the alleviation of suffering and Hardin's objection to Singer.
 - 4. Transition from utilitarianism to Susan James.
- B. Susan James' concept of duty towards those who are

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**Edward P. Bailey, Philip A. Powell, Jack
M. Shuttleworth**



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Reading and Writing for Civic Literacy Donald Lazere, 2015-12-03 This brief edition of a groundbreaking textbook addresses the need for college students to develop critical reading writing and thinking skills for self defense in the contentious arena of American civic rhetoric Designed for first year or more advanced composition and critical thinking courses it is one third shorter than the original edition more affordable for students and easier for teachers to cover in a semester or quarter It incorporates up to date new readings and analysis of controversies like the growing inequality of wealth in America and the debates in the 2008 presidential campaign expressed in opposing viewpoints from the political left and right Exercises help students understand the ideological positions and rhetorical patterns that underlie such opposing views Widely debated issues of whether objectivity is possible and whether there is a liberal or conservative bias in news and entertainment media as well as in education itself are foregrounded as topics for rhetorical analysis

Writing the Annotated Bibliography Luke Beatty, Cynthia Cochran, 2020-06-10 This comprehensive and practical guide covers the elements style and use of annotated bibliographies in the research and writing process for any discipline key disciplinary conventions and tips for working with digital sources Written jointly by a library director and a writing center director this book is packed with examples of individual bibliography entries and full bibliography formats for a wide range of academic needs Online resources include sample bibliographies relevant web links printable versions of checklists and figures and further resources for instructors and researchers Writing the Annotated Bibliography is an essential resource for first year and advanced composition classes courses in writing across the disciplines graduate programs library science instruction programs and academic libraries at the secondary level and beyond It is suitable for both undergraduate and graduate students and for researchers at all levels

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web 1 define each remember oil rig oxidation is loss of e reduction is gain of e a oxidation b reduction c oxidizing agent d reducing agent 2 write half reactions for each of the following atoms or ions label each as oxidation or reduction use the activity series chart a al b ba2 c br 2 d ca e ga3 f h 2 g h

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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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web you should try to answer the questions without referring to your textbook if you get stuck try asking another group for help 1 identify the oxidation state of each element in the following a na 2 so 4 b mno 4 c kclo 3 d ch 3 oh 2 identify the oxidation state of nitrogen in the following a no 3 b no 2 c no 2

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web an oxidation reduction redox reaction is one in which one or more atoms change oxidation numbers oxidation occurs when an atom s oxidation state becomes more positive indicating that electrons have been lost reduction occurs when an atom s oxidation state becomes less positive indicating that electrons have been gained

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web describe redox reactions in terms of electron transfer recall that oxidation involves loss of electrons and reduction involves gain of electrons represent simple redox reactions using half equations identify oxidising and reducing agents for simple redox reactions understand the reactivity of metals and metal displacement reaction

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23 fe2 co co2 fe oxidation co co2 2e reduction fe2 2e fe 24 3 ag ni ni3 3 ag oxidation reduction 25 cu2 pb pb2 cu oxidation reduction 26 o 2

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent

$$\text{a} \text{ i} 2 \text{ o} 5 \text{ 5co i} 2 \text{ o} 5 \text{ 5 c o i} 2 \text{ 5 c o} 2$$

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also

$$2\text{sr} \text{ o} 2 \text{ 2sro sr} 0 \text{ to sr} 2 \text{ oxidized reducing agent o} 0 \text{ to o} 2 \text{ reduced ox ag } 2\text{li s li} 2\text{s li} 0 \text{ to li} 1 \text{ oxidized red ag } \text{s} 0 \text{ to s} 2 \text{ reduced ox ag } 3 \text{ 2cs br} 2$$