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web this experimental approach required the introduction of a new unit for amount of substances the mole which remains

indispensable in modern chemical science the mole is an amount unit similar to familiar units like pair dozen gross etc it provides a specific measure of the number of atoms or molecules in a bulk sample of matter

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web ans the mole concept in chemistry is a fundamental concept that allows us to connect the mass of a substance to the

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number of particles atoms molecules or ions it contains it is based on avogadro s number which states that one mole of any substance contains 6 022 x 10 23 particles 2