



Physical Science

PDF Study Guide

Properties of Elements
 Study Guide: The Periodic Table, Ions, and...
 Questions and Answers
 1. Q: What are the main characteristics of metals?
 A: High electrical and thermal conductivity, and a shiny appearance.
 2. Q: Where are non-metals located on the periodic table?
 A: On the right side of the periodic table.
 3. Q: What is a noble gas?
 A: A gas that is chemically inert.
 4. Q: Name a characteristic of noble gases.
 A: They are inert and rarely form compounds.
 5. Q: What type of elements are found in the middle of the periodic table?
 A: Metals.
 6. Q: What type of elements are found on the far left of the periodic table?
 A: Alkali metals.
 7. Q: What type of elements are found on the far right of the periodic table?
 A: Noble gases.

Properties of Elements
 Study Guide: The Periodic Table, Ions, and Valence Electrons
 1. Overview of the Periodic Table
 The periodic table is a systematic arrangement of elements based on their atomic number, electron configurations, and recurring chemical properties. Elements are organized into horizontal rows (periods) and vertical columns (groups).
 2. Metals, Non-Metals, and Metalloids
 Metals are located on the left side of the periodic table. They are characterized by high electrical and thermal conductivity, malleability, and a shiny appearance. Examples include iron, copper, and aluminum.
 Non-metals are located on the right side of the periodic table. They are generally poor conductors of heat and electricity. Examples include carbon, oxygen, and nitrogen.
 Metalloids are located between metals and non-metals. They have properties intermediate between metals and non-metals. Examples include silicon, boron, and arsenic.

Properties of Elements
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 3. Metals, Non-Metals, and Metalloids (continued)
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