

Class 12

SECTION 3:

NOTE: Answer these short-answer questions in the space provided.

1. Illustration below shows pistons of a fixed mass of gas with different charges, at rest, balanced in the following conditions:

Piston 1	at atmospheric pressure, volume kept constant
Piston 2	below atmospheric, temperature same as the water
Piston 3	at atmospheric pressure, volume kept the same
Piston 4	above atmospheric, temperature same as the water

2. Two sealed flasks, A and B, contain gas at different pressures and volumes at the same temperature. Assume that flask A is connected to flask B by a tube. With the piston in the two flasks volume equal? If not, what is flask with lower the higher pressure? Will the piston will not move at all? If all other factors remain constant, flask A will have higher pressure because it is at a higher temperature.

Circle which is the answer and the flask that will have higher gas pressure in the space provided.

3. A large flask is filled with air at 10°C. Assume that the volume of the gas does not change appreciably when the flask is cooled.

a. If the cooled the air inside the flask at 10°C by cooling it down, does the pressure in the flask increase or decrease? Assume the volume of the flask remains constant, answer if temperature increase or decrease then:

b. Assume the temperature has doubled, does the pressure double? Explain.

c. Assume the pressure has doubled, does the temperature double? Explain. (2 pts)

Review Gases Section 2 Short Answer

S Ashworth



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