

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	58	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

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Tobias Bleicker



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Table of Contents Review Chemistry Answers With Work

1. Understanding the eBook Review Chemistry Answers With Work
 - The Rise of Digital Reading Review Chemistry Answers With Work
 - Advantages of eBooks Over Traditional Books
2. Identifying Review Chemistry Answers With Work
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Review Chemistry Answers With Work
 - User-Friendly Interface
4. Exploring eBook Recommendations from Review Chemistry Answers With Work

- Personalized Recommendations
 - Review Chemistry Answers With Work User Reviews and Ratings
 - Review Chemistry Answers With Work and Bestseller Lists
5. Accessing Review Chemistry Answers With Work Free and Paid eBooks
 - Review Chemistry Answers With Work Public Domain eBooks
 - Review Chemistry Answers With Work eBook Subscription Services
 - Review Chemistry Answers With Work Budget-Friendly Options
 6. Navigating Review Chemistry Answers With Work eBook Formats
 - ePub, PDF, MOBI, and More
 - Review Chemistry Answers With Work Compatibility with Devices
 - Review Chemistry Answers With Work Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review Chemistry Answers With Work
 - Highlighting and Note-Taking Review Chemistry Answers With Work
 - Interactive Elements Review Chemistry Answers With Work
 8. Staying Engaged with Review Chemistry Answers With Work
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review Chemistry Answers With Work
 9. Balancing eBooks and Physical Books Review Chemistry Answers With Work
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Review Chemistry Answers With Work
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Review Chemistry Answers With Work
 - Setting Reading Goals Review Chemistry Answers With Work
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Review Chemistry Answers With Work

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 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
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