

Hydrocarbons – Physical and Chemical Properties

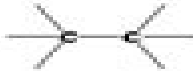
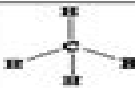
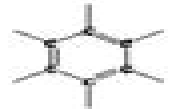
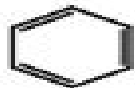
Objective:

In this Lab, we will investigate some of the chemical and physical properties of hydrocarbons and their derivatives and use their properties to distinguish between: alkanes, alkenes, and aromatics

Discussion:

Hydrocarbons are the simplest type of organic compound and they contain only carbon and hydrogen atoms in their structures. Hydrocarbons are classified into different groups based on their chemical bonding pattern. For example: hydrocarbons with only single bonds are known as saturated hydrocarbons (or alkanes) and the ones with multiple bonds (double and triple bonds) are known as unsaturated hydrocarbons (alkenes or alkynes). The detailed classification of saturated and unsaturated hydrocarbons is given in the table below.

Types of hydrocarbons:

Type	Characteristic Functional Group	Simplest Member
Alkane		 Methane (Line-Angle not possible)
Cycloalkane (ring of carbons)		 Cyclopropane
Alkene (has double bond)		 Ethene
Alkyne (has triple bond)		 Ethyne
Aromatic (several double bonds in a ring)		 Benzene

Review Carbon And Hydrocarbons Section 1 Answers

Michael Seilmaier



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Review Carbon And Hydrocarbons Section 1 Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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