

Name _____

Date _____

TYPES OF CHEMICAL REACTIONS

Identify the type of chemical reaction (synthesis, decomposition, single displacement or replacement, double displacement or replacement).

1	$AB + C \rightarrow AC + B$	Single replacement
2	$AB + CD \rightarrow AD + CB$	Double replacement
3	$AB \rightarrow A + B$	Decomposition
4	$A + B \rightarrow AB$	Synthesis
5		Decomposition
6		Double replacement
7		Single replacement
8		Synthesis
9	$\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$	Decomposition
10	$\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$	Double replacement
11	$\text{Zn} + 2\text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$	Single replacement
12	$\text{Na(s)} + \text{Cl}_2\text{(g)} \rightarrow 2\text{NaCl(s)}$	Synthesis
13	$\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$	Double replacement
14	$2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$	Decomposition
15	$\text{Zn(s)} + \text{CuSO}_4\text{(aq)} \rightarrow \text{ZnSO}_4\text{(aq)} + \text{Cu(s)}$	Single replacement

Review And Reinforcement Types Of Chemical Reactions

Answer Key

LL Leslie



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