

## 7-1 Review and Reinforcement

### Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative  
positive  
crisscross

noble gas  
electrons  
anion

octet  
ions  
cation

electron configuration  
polyatomic  
empirical formula

1. An ionic bond is an attraction between oppositely charged ions.
2. Anions have a negative charge.
3. An atom becomes an ion by losing or gaining electrons.
4. The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
6. The atoms that make up polyatomic ions are bonded together by covalent bonds.
7. The crisscross method can be used to write the formula for an ionic compound.
8. The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

losing

true

true

polyatomic

seven

true (released)

9. Calcium becomes a monatomic cation by gaining two electrons.
10. A cation has a positive charge.
11. A binary ionic compound contains only one kind of cation and one kind of anion.
12. Monatomic ions consist of more than one atom.
13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.
14. A great deal of energy is produced when an ionic compound is formed.

# Review And Reinforcement Answers Prentice Hall

**VM Jensen**



## **Review And Reinforcement Answers Prentice Hall:**

## The Enigmatic Realm of **Review And Reinforcement Answers Prentice Hall**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Review And Reinforcement Answers Prentice Hall** a literary masterpiece penned by way of a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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