



## GRADE 10 PHYSICAL SCIENCES

### NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

#### SECTION A

##### QUESTION 1

Various options are provided as possible answers to the following questions. Choose the correct answer and write ONLY the LETTER (A-D) next to the QUESTION/NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A distilled water
- B diamond
- C sodium chloride
- D syngas

1.2 Element with the electron configuration  $1s^2, 2s^2, 2p^4$

- A oxygen
- B carbon
- C helium
- D nitrogen

1.3 The chemical formula of sodium sulfide

- A NaS
- B  $Na_2S$
- C  $NaS_2$
- D  $Na_2S_2$

1.4 Strongest intermolecular force found in water?

- A ion-dipole
- B Dipole-dipole
- C Van der Waals
- D Hydrogen bond

1.5 The particles of a gas ...

- A are tightly packed
- B vibrate against each other
- C move freely and rarely touch
- D move over one another

1.6 Element with the highest electronegativity?

- A Hydrogen
- B Helium
- C Oxygen
- D Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A mass number of the atom will increase
- B atomic number of the atom will decrease
- C number of electrons will decrease
- D number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a

|             |    |
|-------------|----|
| $1s^2$      | 2s |
| $2s^2 2p^4$ | 3s |
| $3s^2$      | 3p |
| $3d^5$      | 4s |

- A noble gas
- B halogen
- C alkali earth metal
- D alkali metal

# Physical Science November Grpaper2 2013

**S Marginson**



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