

Chapter 9 Physical Science Study Guide:

- Know all vocab (on Quizlet)
- Know/describe the different parts of an atom
- Describe the atomic model timeline
 - Scientists, theories, discoveries, models, experiments
- Know what quarks are
- Know what happens to a neutral atom when the number of protons, neutrons, or electrons change
- Be able to describe the three types of nuclear decay

Different Parts of an Atom:

- **Electrons** are very small particles inside an atom with one negative charge (-1). The closer electrons are to the nucleus, less energy.
- The **nucleus** is a small area in the center of the atom where most of its mass and positive charge is concentrated
- A **proton** is an atomic particle that has one positive charge (+1)
- A **neutron** is a neutral particle that exists in the nucleus of an atom
- An **electron cloud** is an area around an atomic nucleus where an electron is most likely to be located
- A **quark** is the smallest particle in an atom, and it is what protons and neutrons are made of

The Atomic Model Timeline:

DEMOCRITUS:

- He was a philosopher who challenged the popular idea of matter.
- His atomic theories were that atoms are small solid objects that cannot be divided, created, or destroyed. Also, that atoms are constantly moving in empty space, and that different types of matter are made of different types of atoms. His last theory is that the properties of the atoms determine the properties of matter.

DALTON:

- In the late 1700s, John Dalton revisited the idea of atoms.
- His atomic theories were that all matter is made of atoms that can't be divided, created, or destroyed. Also, that during a chemical reaction, atoms of one element can't be converted into atoms of another element. The third theory is that atoms of one element are identical to each other but different from atoms of another element. The last theory is that atoms combine in specific ratios.

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SA Adler



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Table of Contents Physical Science Chapter 1study Guide

1. Understanding the eBook Physical Science Chapter 1study Guide
 - The Rise of Digital Reading Physical Science Chapter 1study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chapter 1study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chapter 1study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chapter 1study Guide
 - Personalized Recommendations
 - Physical Science Chapter 1study Guide User Reviews and Ratings
 - Physical Science Chapter 1study Guide and Bestseller Lists
5. Accessing Physical Science Chapter 1study Guide Free and Paid eBooks
 - Physical Science Chapter 1study Guide Public Domain eBooks
 - Physical Science Chapter 1study Guide eBook Subscription Services

- Physical Science Chapter 1study Guide Budget-Friendly Options
- 6. Navigating Physical Science Chapter 1study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Chapter 1study Guide Compatibility with Devices
 - Physical Science Chapter 1study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Chapter 1study Guide
 - Highlighting and Note-Taking Physical Science Chapter 1study Guide
 - Interactive Elements Physical Science Chapter 1study Guide
- 8. Staying Engaged with Physical Science Chapter 1study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chapter 1study Guide
- 9. Balancing eBooks and Physical Books Physical Science Chapter 1study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chapter 1study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Chapter 1study Guide
 - Setting Reading Goals Physical Science Chapter 1study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Chapter 1study Guide
 - Fact-Checking eBook Content of Physical Science Chapter 1study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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