

**GRADE 11**

**PHYSICAL SCIENCES**

**P1 (PHYSICS)**

**FINAL/NOVEMBER**

**EXAM**

**WITH  
MEMORANDUM**

# Physical Science Grade 11 November 2014

**Xiaolong Qi**



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web the molar volume will be the volume occupied by one molar mass 32 g of the liquid expressing the density in liters instead of ml we have  $v_m = \frac{32 \text{ g mol}^{-1}}{1790 \text{ g l}^{-1}} = 0.0405 \text{ l mol}^{-1}$

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web the molar volume has the si unit of cubic metres per mole  $\text{m}^3 \text{mol}^{-1}$  although it is more typical to use the units cubic decimetres per mole  $\text{dm}^3 \text{mol}^{-1}$  for gases and cubic centimetres per mole  $\text{cm}^3 \text{mol}^{-1}$  for liquids and solids

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