



MERCY JUNIOR COLLEGE  
Grade 11  
Physical Science  
Finals Exam

NAME: \_\_\_\_\_

YEAR LEVEL & SECTION: \_\_\_\_\_

TEACHER'S NAME: \_\_\_\_\_

PROCTOR'S NAME: \_\_\_\_\_

1. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol  $e^-$  or  $\beta^-$ , with a negative elementary electric charge.
  - Neutron
  - Electron
  - Proton
  - Atom
- It is a subatomic particle, symbol  $p$  or  $p^+$ , with a positive electric charge of  $+1e$  elementary charge and mass slightly less than that of a neutron.
  - Neutron
  - Electron
  - Proton
  - Atom
- It is a subatomic particle, symbol  $n$  or  $n^0$ , with no net electric charge and a mass slightly larger than that of a proton.
  - Neutron
  - Electron
  - Proton
  - Atom
- Is this an example of ionic bonding?



carbon dioxide

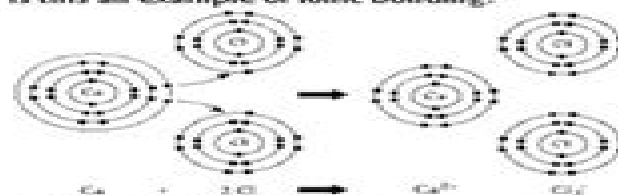
- Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.

- Is this an example of ionic bonding?



lithium fluoride

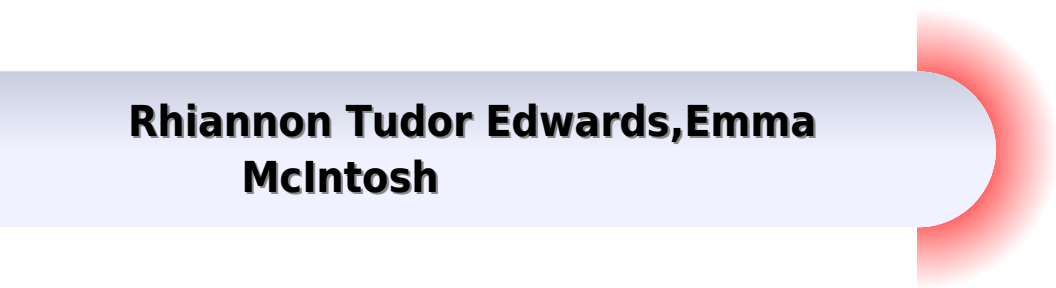
- Yes, it is ionic bonding.
  - No, it is of another type of chemical bonding.
  - There is no chemical bonding present.
  - This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
    - Electron Configuration
    - Periodic Table of Elements
    - Orbital Models
    - Energy Levels
  - Is this an example of ionic bonding?



- Yes, it is ionic bonding.
  - No, it is of another type of chemical bonding.
  - There is no chemical bonding present.
  - This is just an orbital model.
- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
    - Octet Rule
    - Valence Electrons
    - Atomic Orbitals
    - Chemical Bonds
  - It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
    - Neutron
    - Electron
    - Proton
    - Atom
  - It is a type of chemical bond that involves the electrostatic attraction between oppositely

# Physical Science Grade 11 Final Examination 2014 Scope

**Rhiannon Tudor Edwards, Emma  
McIntosh**



**Physical Science Grade 11 Final Examination 2014 Scope:**

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In a world eaten by screens and the ceaseless chatter of instantaneous conversation, the melodic splendor and emotional symphony created by the prepared term frequently fade into the backdrop, eclipsed by the persistent noise and distractions that permeate our lives. However, located within the pages of **Physical Science Grade 11 Final Examination 2014 Scope** a wonderful literary treasure filled with natural emotions, lies an immersive symphony waiting to be embraced. Crafted by a wonderful composer of language, that fascinating masterpiece conducts readers on an emotional journey, skillfully unraveling the hidden songs and profound influence resonating within each carefully crafted phrase. Within the depths with this emotional analysis, we shall discover the book is central harmonies, analyze its enthralling publishing fashion, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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