

- 10. The reaction is exothermic because the products have a lower enthalpy than the reactants.
- 11. The reaction is exothermic because the products have a lower enthalpy than the reactants.
- 12. The reaction is exothermic because the products have a lower enthalpy than the reactants.
- 13. The reaction is exothermic because the products have a lower enthalpy than the reactants.



Figure 10



Figure 11

Figure 10 shows the enthalpy change for the reaction of hydrogen and oxygen to form water. The reaction is exothermic because the products have a lower enthalpy than the reactants.

Answer the questions in the margin.

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