

2019 Operational calendar

1440-1441

إرامكو السعودية
saudi aramco



January

1440: Rabi' al Thani (29)
Jumada I (29)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
		1	2	3	4	5
6	7	8	9	10	11	12
13	14	15	16	17	18	19
20	21	22	23	24	25	26
27	28	29	30	31		

بداية شهر ربيع الثاني 1440 هـ

February

Jumada I (29)
Jumada II (29)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
					1	2
3	4	5	6	7	8	9
10	11	12	13	14	15	16
17	18	19	20	21	22	23
24	25	26	27	28		

March

Jumada II (29)
Rajab (27)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
					1	2
3	4	5	6	7	8	9
10	11	12	13	14	15	16
17	18	19	20	21	22	23
24	25	26	27	28	29	30
31						

April

Rajab (27)
Shawwal (30)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

بداية شهر ربيع الثاني 1440 هـ

May

Shawwal (30)
Ramadan (30)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
			1	2	3	4
5	6	7	8	9	10	11
12	13	14	15	16	17	18
19	20	21	22	23	24	25
26	27	28	29	30	31	

بداية شهر ربيع الثاني 1440 هـ

June

Ramadan (30)
Shawwal (30)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

July

Shawwal (30)
Dhu al-Qadah (11)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

August

Dhu al-Qadah (11)
Dhu al-Hijjah (12)
1441: Muharram (30)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

بداية شهر ربيع الثاني 1441 هـ

September

1441: Muharram (30)
Safar (29)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

بداية شهر ربيع الثاني 1441 هـ

October

Safar (29)
Rabi' al Thani (29)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

November

Rabi' al Thani (29)
Rabi' al Thani (29)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

بداية شهر ربيع الثاني 1441 هـ

December

Rabi' al Thani (29)
Jumada I (29)

Sun	Mon	Tue	Wed	Thu	Fri	Sat
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

Holidays

Ramadan

Rescheduled Days-Off

Weekends

National Day

أحمد بن عبد الله العتيق
Ahmed Aldehaim (aldehaim)
Corporate Business Applications Dept.

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N Colangelo



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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an acid and a base react not to produce a salt and a solvent but to form a new acid and a new base the concept of

neutralization is thus absent brønsted lowry acid base behavior is formally independent of any solvent making it more all encompassing than the arrhenius model

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acids bases and enzymes many acids and bases in living things provide the ph that enzymes need enzymes are biological

catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper pH for enzymes to do their work

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

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oct 18 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low pH value i.e. below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

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arrhenius's definition of acids and bases the earliest definition of acids and bases is arrhenius's definition which states that an acid is a substance that forms hydrogen ions H^+ when dissolved in water and a base is a substance that forms hydroxide ions OH^- when dissolved in water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other's chemistry to produce a rather innocuous substance water in fact the

general reaction between an acid and a base is ce acid base water salt nonumber

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