Directions: Complete the table below by writing the name of each of the substances above under the proper heading. Place substances with a pH lower than 3.0 in the strong acids column. Place substances with a pH higher than 10.0 in the strong bases column.

| 1. Strong acids | 2. Weak acids | 3. Weak bases | 4. Strong bases |
|-----------------|---------------|---------------|-----------------|
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Directions: Answer the following questions on the lines provided.

- Is pure water an acidic, basic, or neutral substance? neutral
- 6. Is the pH of a strong acid higher or lower than the pH of a weak acid of the same concentration?
 The pH of a strong acid is lower than that of a weak acid.
- 7. Is the pH of a strong base higher or lower than the pH of a weak base of the same concentration? The pH of a strong base is higher than that of a weak base.
- 8. On the pH scale, what are the values of acids and what are the values of bases?
- The pH of acids is less than 7; the pH of a base is greater than 7.





Review And Reinforcement Answer Key Acids

JR Anderson

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