- Given the nuclear equation: $147N + X \rightarrow 162O + 21H$ What is particle X7 A) an alpha particle B) a beta particle. C) a deuteron. D) a triton.
- 2. The nucleus of a radium-226 atom is unstable, which causes the nucleus to spontaneously
 - A) absorb electrons
- B) absorb protons

C) decay

- D) oxidize
- Which group of nuclear emissions is listed in order of increasing charge?
 - A) alpha particle, beta particle, gamma radiation =
 - B) gamma radiation, alpha particle, beta particle —
 - C) positron, alpha particle, neutron
 - D) neutron, positron, alpha particle
- 4. Which list of nuclear emissions is arranged in order from the least penetrating power to the greatest penetrating p-owner?
 - A) alpha particle, beta particle, gamma ray
 - B) alpha particle, gamma ray, beta particle
 - C) gamma ray, beta particle, alpha particle
 - D) beta particle, alpha particle, gamma ray
- 5. Base your answer to the following question on Given the nuclear equation:

$$^{1}iH + X \rightarrow ^{6}iLi + ^{4}He$$

The particle represented by X is

- A) 94Li /B) 94Be C) 105Be D) 106C

- 6. Positrons are spontaneously emitted from the nuclei of
 - A) potassium-37
- B) radium-226
- C) nitrogen-16
- D) thorium-232
- 7. What is the decay mode of 37K?
 - A) 3-
- (c) (7)
- DO a
- 8. Which equation represents positron decay?
 - A) 27 77 Rb -> 0 -1 e + 27 76 Sr
 - B) $277 \odot U \rightarrow 223 \odot Th + 40 He$
 - (c) $^{27}_{13}Al + ^{4}_{2}He \rightarrow ^{30}_{15}P + ^{1}_{9}n$
 - D) 116 C 0+1e+115B

- 9. As ¹⁴C decays to ¹⁴N, the number of protons in the nucleus.
 - A) decreases

- B) increases
- C) remains the same
- 10. Atoms of I-131 spontaneously decay when the
 - A) stable nuclei emit alpha particles.
 - B) stable nuclei emit beta particles
 - C) unstable nuclei emit alpha particles
 - D) unstable nuclei emit beta particles
- 11. Which statement best describes gamma radiation?
 - A) It has a mass of 1 and a charge of 1.
 - B) It has a mass of 0 and a charge of -1.
 - (C) It has a mass of 0 and a charge of 0.
 - D) It has a mass of 4 and a charge of +2.
- 12. Which nuclear decay emission consists of energy, only?
 - A) alpha particle
- B) beta particle.
- C) gamma radiation
- D)) positron
- Which kind of nuclear radiation has high energy and no 1008.557
 - A) alpha

B) beta

C) gamma

- D) neutron
- The half-life of a radioactive isotope is 20.0 minutes. What is the total amount of a 1.00-gram sample of this isotope remaining after 1.00 hour?
 - A) 0.500 g

B) 0.333 g

C) 0.250 g

- D) 0.125 g
- 15. A radioactive element has a half-life of 2 days. Which fraction represents the amount of an original sample of this element remaining after 6 days?



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- 1. Understanding the eBook Review Nuclear Chemistry Answers
 - The Rise of Digital Reading Review Nuclear Chemistry Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Review Nuclear Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Review Nuclear Chemistry Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Review Nuclear Chemistry Answers
 - Personalized Recommendations
 - Review Nuclear Chemistry Answers User Reviews and Ratings

- Review Nuclear Chemistry Answers and Bestseller Lists
- 5. Accessing Review Nuclear Chemistry Answers Free and Paid eBooks
 - Review Nuclear Chemistry Answers Public Domain eBooks
 - Review Nuclear Chemistry Answers eBook Subscription Services
 - Review Nuclear Chemistry Answers Budget-Friendly Options
- 6. Navigating Review Nuclear Chemistry Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Review Nuclear Chemistry Answers Compatibility with Devices
 - Review Nuclear Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review Nuclear Chemistry Answers
 - Highlighting and Note-Taking Review Nuclear Chemistry Answers
 - Interactive Elements Review Nuclear Chemistry Answers
- 8. Staying Engaged with Review Nuclear Chemistry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review Nuclear Chemistry Answers
- 9. Balancing eBooks and Physical Books Review Nuclear Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Review Nuclear Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Review Nuclear Chemistry Answers
 - Setting Reading Goals Review Nuclear Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review Nuclear Chemistry Answers
 - Fact-Checking eBook Content of Review Nuclear Chemistry Answers
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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