

Unit 8 Review

Write *increases* or *decreases* to complete each relationship.

1. As volume increases, pressure _____decreases_____.
2. As volume increases, temperature _____increases_____.
3. As pressure increases, temperature _____increases_____.
4. As pressure increases, concentration _____increases_____.
5. As force decreases, pressure _____decreases_____.

True or False.

6. ____T_____ The Kinetic Theory of Matter deals with moving particles.
7. ____T_____ Gases can be collected by trapping them over water.
8. ____T_____ Real gases exist in the world around us.
9. ____T_____ Densities of gases are lower than 1 g/mL.
10. ____F_____ If the barometric pressure is falling, that means good weather is coming.
11. ____T_____ The troposphere contains the most air particles.
12. ____T_____ Temperature is a measure of how fast particles are moving.

13. The gas that makes up the highest percentage of the atmosphere is

____nitrogen_____.

14. Atmospheric pressure is measured using a ____barometer_____.

15. Squeezing gas particles together is known as ____compressibility_____.

16. When particles collide with the sides of a container, they create ____pressure_____.

17. The spreading of a gas through a small opening is ____effusion_____.

18. The number of gas particles present is called the ____concentration_____.

Gas Law Problems. Write the answer with the correct unit.

19. A gas has a beginning pressure of 2.95 atm and a volume of 5.15 L. If the ending pressure is 4.23 atm, what is the ending volume? $P_1V_1 = P_2V_2$
 $(2.95)(5.15) \div 4.23 = 3.59$

20. A gas has a beginning temperature of 300 K and a volume of 4.06 L. If the ending temperature is 176 K, what is the ending volume? $V_1T_2 = V_2T_1$
 $(4.06)(176) \div 300 = 2.38$

21. A gas has a beginning pressure of 2.09 atm, a temperature of 268 K, and a volume of 0.450 L. If the ending pressure is 1.2 atm and the temperature is 149 K, what is the ending volume? $P_1V_1T_2 = P_2V_2T_1$

Review Gases Section Answers

N Colangelo



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Table of Contents Review Gases Section Answers

1. Understanding the eBook Review Gases Section Answers
 - The Rise of Digital Reading Review Gases Section Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Review Gases Section Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Review Gases Section Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Review Gases Section Answers
 - Personalized Recommendations
 - Review Gases Section Answers User Reviews and Ratings
 - Review Gases Section Answers and Bestseller Lists
5. Accessing Review Gases Section Answers Free and Paid eBooks
 - Review Gases Section Answers Public Domain eBooks
 - Review Gases Section Answers eBook Subscription Services
 - Review Gases Section Answers Budget-Friendly Options
6. Navigating Review Gases Section Answers eBook Formats

- ePub, PDF, MOBI, and More
- Review Gases Section Answers Compatibility with Devices
- Review Gases Section Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review Gases Section Answers
 - Highlighting and Note-Taking Review Gases Section Answers
 - Interactive Elements Review Gases Section Answers
- 8. Staying Engaged with Review Gases Section Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review Gases Section Answers
- 9. Balancing eBooks and Physical Books Review Gases Section Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Review Gases Section Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Review Gases Section Answers
 - Setting Reading Goals Review Gases Section Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review Gases Section Answers
 - Fact-Checking eBook Content of Review Gases Section Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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