

Chemical Reactions Review

Explore the difference between the vocabulary terms in each of the following sets.

1. exothermic reaction – endothermic reaction

exo – releases heat energy during the reaction

endo – absorbs heat energy, feels cool to the touch

2. reactant – product

reactant – substances used in a reaction

product – substances produced from a chemical reaction

3. chemical equation – reaction

chemical equation – ways of describing a chemical reaction using chemical formulas

reaction – substances used in a reaction

Circle the word or phrase that best answers the question.

4. Which of these is a chemical change?

a) Paper is shredded.

b) Liquid wax turns solid.

c) A raw egg is broken.

d) Wood chips burn.

5. Which of these reactions releases heat energy?

a) Unbalanced

b) Cooked

c) Exothermic

d) Endothermic

6. Which is NOT evidence that a chemical reaction has occurred?

a) Milk turns sour.

b) Steam condenses on a cold window.

c) A strong odor comes from a broken egg.

d) A slice of raw potato browns.

7. Jeff was trying to determine if his experiment had undergone a physical or chemical change. Which observation could mean that it was NOT a chemical change?

a) Something was burning.

c) Gas and bubbles were created.

b) The temperature changed.

d) The colors of the reactants mixed.

8. Evidence that a chemical change has occurred includes a _____.

a) Change in mass

b) Change in size

c) Change in shape

d) Change in color

Review Chemical Reactions Answer

LL Leslie



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