

# POLYATOMIC IONS

## Ions with -1 charge

perbromate	$\text{BrO}_4^{-1}$
bromate	$\text{BrO}_3^{-1}$
bromite	$\text{BrO}_2^{-1}$
hypobromite	$\text{BrO}^{-1}$
perchlorate	$\text{ClO}_4^{-1}$
chlorate	$\text{ClO}_3^{-1}$
chlorite	$\text{ClO}_2^{-1}$
hypochlorite	$\text{ClO}^{-1}$
periodate	$\text{IO}_4^{-1}$
iodate	$\text{IO}_3^{-1}$
iodite	$\text{IO}_2^{-1}$
hypoiodite	$\text{IO}^{-1}$

nitrate	$\text{NO}_3^{-1}$
nitrite	$\text{NO}_2^{-1}$
hydroxide	$\text{OH}^{-1}$
cyanide	$\text{CN}^{-1}$
thiocyanate	$\text{SCN}^{-1}$
acetate	$\text{C}_2\text{H}_3\text{O}_2^{-1}$
permanganate	$\text{MnO}_4^{-1}$
bicarbonate	$\text{HCO}_3^{-1}$

## Ions with a -2 Charge

carbonate	$\text{CO}_3^{-2}$
phthalate	$\text{C}_8\text{H}_4\text{O}_4^{-2}$
sulfate	$\text{SO}_4^{-2}$

sulfite	$\text{SO}_3^{-2}$
chromate	$\text{CrO}_4^{-2}$
dichromate	$\text{Cr}_2\text{O}_7^{-2}$
oxalate	$\text{C}_2\text{O}_4^{-2}$
peroxide	$\text{O}_2^{-2}$

## Ions with a -3 Charge

phosphate	$\text{PO}_4^{-3}$
phosphite	$\text{PO}_3^{-3}$
arsenate	$\text{AsO}_4^{-3}$

## Ions with +1 charge

ammonium ion	$\text{NH}_4^{+1}$
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# Polyatomic Ion Packet Review

**Judd E. Hollander**



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## **Table of Contents Polyatomic Ion Packet Review**

1. Understanding the eBook Polyatomic Ion Packet Review
  - The Rise of Digital Reading Polyatomic Ion Packet Review
  - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ion Packet Review
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Polyatomic Ion Packet Review
  - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ion Packet Review
  - Personalized Recommendations
  - Polyatomic Ion Packet Review User Reviews and Ratings
  - Polyatomic Ion Packet Review and Bestseller Lists
5. Accessing Polyatomic Ion Packet Review Free and Paid eBooks
  - Polyatomic Ion Packet Review Public Domain eBooks
  - Polyatomic Ion Packet Review eBook Subscription Services
  - Polyatomic Ion Packet Review Budget-Friendly Options
6. Navigating Polyatomic Ion Packet Review eBook Formats

- ePub, PDF, MOBI, and More
- Polyatomic Ion Packet Review Compatibility with Devices
- Polyatomic Ion Packet Review Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Polyatomic Ion Packet Review
  - Highlighting and Note-Taking Polyatomic Ion Packet Review
  - Interactive Elements Polyatomic Ion Packet Review
- 8. Staying Engaged with Polyatomic Ion Packet Review
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Polyatomic Ion Packet Review
- 9. Balancing eBooks and Physical Books Polyatomic Ion Packet Review
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Polyatomic Ion Packet Review
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Polyatomic Ion Packet Review
  - Setting Reading Goals Polyatomic Ion Packet Review
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Polyatomic Ion Packet Review
  - Fact-Checking eBook Content of Polyatomic Ion Packet Review
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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