

Polyatomic Ion Worksheet

Place the charges on each ion – Write the Formula – Name the Compound

Remember – The cation always goes first.

Example:	NH_4^{+1}	&	I^{-1}	Formula:	<u>NH_4I</u>	Name:	<u>Ammonium Iodide</u>
1.	NH_4	&	O	Formula:	_____	Name:	_____
2.	Ca	&	CO_3	Formula:	_____	Name:	_____
3.	HCO_3	&	Na	Formula:	_____	Name:	_____
4.	NO_3	&	Mg	Formula:	_____	Name:	_____
5.	K	&	SO_4	Formula:	_____	Name:	_____
6.	Li	&	PO_4	Formula:	_____	Name:	_____

Finding the Formula from the Name

Example:	Lithium Chloride	Ions:	<u>Li^{+1} & Cl^{-1}</u>	Formula:	<u>LiCl</u>
1.	Sodium Oxide	Ions:	_____	Formula:	_____
2.	Potassium Bromide	Ions:	_____	Formula:	_____
3.	Calcium Nitride	Ions:	_____	Formula:	_____
4.	Beryllium Sulfide	Ions:	_____	Formula:	_____
5.	Magnesium Iodide	Ions:	_____	Formula:	_____
6.	Aluminum Oxide	Ions:	_____	Formula:	_____
7.	Ammonium Bromide	Ions:	_____	Formula:	_____
8.	Calcium Bicarbonate	Ions:	_____	Formula:	_____
9.	Potassium Phosphate	Ions:	_____	Formula:	_____
10.	Aluminum Sulfate	Ions:	_____	Formula:	_____

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Brendan G. Carr



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Table of Contents Polyatomic Ions Worksheet With Answers Physical Science If8767

1. Understanding the eBook Polyatomic Ions Worksheet With Answers Physical Science If8767
 - The Rise of Digital Reading Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Polyatomic Ions Worksheet With Answers Physical Science If8767
 - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Personalized Recommendations
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 User Reviews and Ratings
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 and Bestseller Lists
5. Accessing Polyatomic Ions Worksheet With Answers Physical Science If8767 Free and Paid eBooks
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Public Domain eBooks
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 eBook Subscription Services
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Budget-Friendly Options

6. Navigating Polyatomic Ions Worksheet With Answers Physical Science If8767 eBook Formats
 - ePub, PDF, MOBI, and More
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Compatibility with Devices
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Highlighting and Note-Taking Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Interactive Elements Polyatomic Ions Worksheet With Answers Physical Science If8767
8. Staying Engaged with Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polyatomic Ions Worksheet With Answers Physical Science If8767
9. Balancing eBooks and Physical Books Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polyatomic Ions Worksheet With Answers Physical Science If8767
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Setting Reading Goals Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Fact-Checking eBook Content of Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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