

A chemical reaction is a transformation of a chemical reaction in which the number of atoms of each element is conserved.

A coefficient represents the number of atoms of a particular element in a chemical reaction. A coefficient is placed directly in front of a chemical formula.

A subscript is placed after the element symbol to indicate the number of atoms of that element in a chemical formula.

Why is it important to balance chemical equations?

Chemical reactions are balanced to show that the number of atoms of each element is the same on both sides of the equation.

What does the law of conservation of mass state?

The law of conservation of mass states that the total mass of the reactants is equal to the total mass of the products.

Week 8, Day 2

The number of protons in an atom cannot change (it is fixed) but a different number of neutrons can change.

The most stable atomic configuration has a nucleus with a balanced number of protons and neutrons.

How does a positive ion form from a neutral atom? What is another name for a positive ion?

When an atom loses one or more electrons, it has more protons than electrons and is no longer neutral. This atom has a positive charge and is called a positive ion.

How does a negative ion form from a neutral atom? What is another name for a negative ion?

When an atom gains one or more electrons, it has more electrons than protons and is no longer neutral. This atom has a negative charge and is called a negative ion.



Atoms in Group 1A have 1 electron in their outer shell and Group 2A atoms have 2 electrons in their outer shell. To become more stable (like the noble gases), they easily give up their valence electrons. Group 1A atoms have 1 electron in their outer shell and easily lose it. Group 2A atoms have 2 electrons in their outer shell and easily lose them.



Atoms are formed by losing the first part of the element name and adding the suffix -ide. For example, sodium chloride, NaCl, is called a sodium ion.



When sodium and chlorine atoms react, the sodium atom gives up the valence electron, becoming a cation (positively charged ion), and the chlorine atom gains the electron, becoming an anion (negatively charged ion).

The positively charged sodium ion and the negatively charged chloride ion are attracted to each other. The bond formed between the positively charged ion and the negatively charged ion is called an ionic bond.

Physical Science Exploring Space Notetaking Guide Answer

Chao Zhang



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Table of Contents Physical Science Exploring Space Notetaking Guide Answer

1. Understanding the eBook Physical Science Exploring Space Notetaking Guide Answer
 - The Rise of Digital Reading Physical Science Exploring Space Notetaking Guide Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Exploring Space Notetaking Guide Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Exploring Space Notetaking Guide Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Exploring Space Notetaking Guide Answer

- Personalized Recommendations
- Physical Science Exploring Space Notetaking Guide Answer User Reviews and Ratings
- Physical Science Exploring Space Notetaking Guide Answer and Bestseller Lists
- 5. Accessing Physical Science Exploring Space Notetaking Guide Answer Free and Paid eBooks
 - Physical Science Exploring Space Notetaking Guide Answer Public Domain eBooks
 - Physical Science Exploring Space Notetaking Guide Answer eBook Subscription Services
 - Physical Science Exploring Space Notetaking Guide Answer Budget-Friendly Options
- 6. Navigating Physical Science Exploring Space Notetaking Guide Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Exploring Space Notetaking Guide Answer Compatibility with Devices
 - Physical Science Exploring Space Notetaking Guide Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Exploring Space Notetaking Guide Answer
 - Highlighting and Note-Taking Physical Science Exploring Space Notetaking Guide Answer
 - Interactive Elements Physical Science Exploring Space Notetaking Guide Answer
- 8. Staying Engaged with Physical Science Exploring Space Notetaking Guide Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Exploring Space Notetaking Guide Answer
- 9. Balancing eBooks and Physical Books Physical Science Exploring Space Notetaking Guide Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Exploring Space Notetaking Guide Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Exploring Space Notetaking Guide Answer
 - Setting Reading Goals Physical Science Exploring Space Notetaking Guide Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Exploring Space Notetaking Guide Answer

- Fact-Checking eBook Content of Physical Science Exploring Space Notetaking Guide Answer
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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