

Physical Science EOCT Study Guide – Domain 3

Physical Energy, Force, and Motion

Standards: PS.3-7 and 8

1. Students will understand relationships and flow of energy within a system.
 - a. Explain thermal energy (heat) and kinetic energy (movement) and how they are related.
 - b. Explain heat. Thermal energy flows from objects that are hotter to objects that are cooler.
 - c. Explain specific heat. The amount of energy needed to raise the temperature of 1 kg of a substance by 1°C.
 - d. What is the formula for determining that change in heat that energy can add?

$$Q = mc\Delta T$$

$$\Delta T = T_2 - T_1$$
 - e. A sample of 0.5 kg of water at 20°C and specific heat of 4186 J/kg°C. Find the change in heat energy if the block is 2 mmol from 10°C to 20°C.

$$Q = mc\Delta T$$

$$Q = (0.5 \text{ kg})(4186 \text{ J/kg°C})(20^\circ\text{C} - 10^\circ\text{C})$$

$$Q = 41,860 \text{ J}$$
 - f. Explain conduction, convection, and radiation. Give examples of each.
 - Conduction: heat transfer by contact. Example: a metal spoon in a pot of soup.
 - Convection: heat transfer by movement of a fluid. Example: a pot of soup being stirred.
 - Radiation: heat transfer by electromagnetic waves. Example: heat from the sun.

2. Explain the following phase change diagrams.



3. Students will determine relationships among force, mass and motion.
 - a. Explain the difference between distance and displacement.
 - Distance is the total path length traveled.
 - Displacement is the straight-line distance from start to end.
 - b. What is the difference between average speed and instantaneous speed?
 - Average speed is the total distance divided by total time.
 - Instantaneous speed is the speed at a specific moment.
 - c. What is the formula for average speed?

$$v = \frac{d}{t}$$

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Physical Science Eoct Study Guide Energy Transformations

Peipei Pang



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Table of Contents Physical Science Eoct Study Guide Energy Transformations

1. Understanding the eBook Physical Science Eoct Study Guide Energy Transformations
 - The Rise of Digital Reading Physical Science Eoct Study Guide Energy Transformations
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Eoct Study Guide Energy Transformations
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Eoct Study Guide Energy Transformations
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Eoct Study Guide Energy Transformations
 - Personalized Recommendations
 - Physical Science Eoct Study Guide Energy Transformations User Reviews and Ratings
 - Physical Science Eoct Study Guide Energy Transformations and Bestseller Lists
5. Accessing Physical Science Eoct Study Guide Energy Transformations Free and Paid eBooks

- Physical Science Eoct Study Guide Energy Transformations Public Domain eBooks
 - Physical Science Eoct Study Guide Energy Transformations eBook Subscription Services
 - Physical Science Eoct Study Guide Energy Transformations Budget-Friendly Options
6. Navigating Physical Science Eoct Study Guide Energy Transformations eBook Formats
- ePub, PDF, MOBI, and More
 - Physical Science Eoct Study Guide Energy Transformations Compatibility with Devices
 - Physical Science Eoct Study Guide Energy Transformations Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Eoct Study Guide Energy Transformations
 - Highlighting and Note-Taking Physical Science Eoct Study Guide Energy Transformations
 - Interactive Elements Physical Science Eoct Study Guide Energy Transformations
8. Staying Engaged with Physical Science Eoct Study Guide Energy Transformations
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Eoct Study Guide Energy Transformations
9. Balancing eBooks and Physical Books Physical Science Eoct Study Guide Energy Transformations
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Eoct Study Guide Energy Transformations
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Eoct Study Guide Energy Transformations
- Setting Reading Goals Physical Science Eoct Study Guide Energy Transformations
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Eoct Study Guide Energy Transformations
- Fact-Checking eBook Content of Physical Science Eoct Study Guide Energy Transformations
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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