

- a) Benzene.
- b) Octane.
- c) Pentane.
- d) Entanol.

ANSWER : Pentane.

39) When a solute is dissolved in water it shows:

- a) decrease in freezing point of water.
- b) decrease in boiling point of water.
- c) increase in vapour pressure of water.
- d) all of the above.

ANSWER : decrease in freezing point of water.

40) When sugar is added to water, what is the change observed in boiling and freezing points of water?

- a) Both boiling point and freezing point decreases.
- b) Both boiling point and freezing point increases.
- c) Boiling point increases and freezing point decreases.
- d) Boiling point decreases and freezing point increases.

ANSWER : Boiling point increases and freezing point decreases.

41) aqueous metal solution have highest freezing point.

- a) Urea
- b) Barium chloride.
- c) Potassium bromide.
- d) Aluminium sulphate.

ANSWER : Urea

42) When a thin slice of sugar beet is placed in a concentrated solution of NaCl then sugar beet will.....

- a) dissolve in solution.
- b) neither absorb nor lose water.
- c) absorb water from solution.
- d) lose water from its cell.

ANSWER : lose water from its cell.

43) The wrong relation between osmotic pressure (P), volume (V) and temperature (T) is:

- a) $P \propto n$ if T and V are constant.
- b) Pv is constant if T is constant.
- c) $P \propto V$ if T is constant.
- d) $P \propto T$ if V is constant.

ANSWER : $P \propto V$ if T is constant.

44) Two solutions C and D are separated by a semi-permeable membrane. If liquid flows from D to C then.

- a) Both have same concentration.
- b) D is less concentrated than C.
- c) D is more concentrated than C.
- d) None of these.

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 - The Rise of Digital Reading Physical Chemistry Mcqs
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Chemistry Mcqs
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an Physical Chemistry Mcqs
 - User-Friendly Interface
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 - Physical Chemistry Mcqs User Reviews and Ratings

- Physical Chemistry Mcqs and Bestseller Lists
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 - Highlighting and Note-Taking Physical Chemistry Mcqs
 - Interactive Elements Physical Chemistry Mcqs
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Chemistry Mcqs
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Chemistry Mcqs
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Chemistry Mcqs
 - Setting Reading Goals Physical Chemistry Mcqs
 - Carving Out Dedicated Reading Time
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 - Fact-Checking eBook Content of Physical Chemistry Mcqs
 - Distinguishing Credible Sources

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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
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