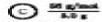
#### Unit 1 Progress Check: MCQ

- 1. Which of the following numerical expressions gives the number of moles in 5.0 g of CaO?
- 5.0 g × 55 g/mol







 In a lab a student is given a 21 g sample of pure Co metal. Which of the following pieces of information is most useful for determining the number of Co atoms in the sample? Assume that the pressure and temperature in the lab are 1.0 atom and 25 °C.

↑ The molar mass of **Cu** 



- n The density of Ou at 25°C
- C The volume of the Cu sample
- (D) The ratio of the two main isotopes found in pure Ou
- A 1.0 mol sample of which of the following compounds has the greatest mass?

# Page 1 Science Lab Progress Check Unit C2 Answers

Jessica J Manson

Page	1	Science	Lab	<b>Progress</b>	Check	Unit	C2	Answer	S:

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