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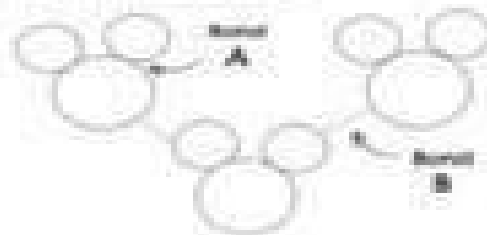
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Review: Properties of Ocean Water

Part I - The Water Molecule

Instructions: The diagram below illustrates how water molecules

1. are made up of two atoms with either one or two electrons each joined together to create the molecule.
2. have a "shape" (geometry) to them that changes if each atom in the molecule
3. identify each molecule that exists.



Identify each molecule that exists.

Identify each molecule that exists.

Part II - Phases of Water

Instructions: For each of the three diagrams below, write the phase of water shown in the top box. Then fill in the rest of the box with information for each phase.



Phase	Diagram	Describe the spacing of molecules in each phase
Solid		
Liquid		
Gas		

Instructions: The diagram below illustrates how water molecules are joined together to create the molecule.

1. Water is made up of two atoms with either one or two electrons each joined together to create the molecule.
2. Water is made up of two atoms with either one or two electrons each joined together to create the molecule.
3. Water is made up of two atoms with either one or two electrons each joined together to create the molecule.
4. Water is made up of two atoms with either one or two electrons each joined together to create the molecule.
5. Water is made up of two atoms with either one or two electrons each joined together to create the molecule.
6. Water is made up of two atoms with either one or two electrons each joined together to create the molecule.
7. Water is made up of two atoms with either one or two electrons each joined together to create the molecule.
8. Water is made up of two atoms with either one or two electrons each joined together to create the molecule.

Part III - Density

Instructions: The diagram below illustrates how water molecules are joined together to create the molecule. The diagram shows four beakers labeled A, B, C, and D. Each beaker contains a different amount of water. The water level in each beaker is indicated by a horizontal line.



Part IV - Density Temperature

1. Density is the mass of a substance divided by its volume.
2. Density is the mass of a substance divided by its volume.



Part V - Density and Volume

Instructions: The diagram below illustrates how water molecules are joined together to create the molecule. The diagram shows four beakers labeled A, B, C, and D. Each beaker contains a different amount of water. The water level in each beaker is indicated by a horizontal line.

1. Density is the mass of a substance divided by its volume.
2. Density is the mass of a substance divided by its volume.
3. Density is the mass of a substance divided by its volume.

Instructions: The diagram below illustrates how water molecules are joined together to create the molecule. The diagram shows four beakers labeled A, B, C, and D. Each beaker contains a different amount of water. The water level in each beaker is indicated by a horizontal line.



Ocean Water Concept Review

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