

Answer Key

Guided Reading 6.1

Observing Chemical Change Guided Reading and Study Use Target Reading Skills Sample questions and answers: Question What are physical properties of matter? Answer Physical properties are characteristics that can be observed without changing one substance into another. Question What is the evidence for chemical reactions? Answer The evidence for chemical reactions is the formation of new substances and changes in energy.

1. Matter is anything that has mass and takes up space.
2. chemistry
3. a. Physical b. Only by changing one substance into another
4. False
5. a, b, d
6. chemical reaction, or chemical change
7. They break and form new bonds, producing new substances.
8. Formation of new substances and changes in energy
9. The change in color is an indication of a change in properties, and a change in properties is a sign that a chemical reaction occurred.
10. precipitate
11. It might indicate a chemical reaction, because the bubbles are gas, and the presence of a gas indicates a new substance may have been formed.
12. True
13. A change in temperature reflects a change in energy.
14. False
15. a. Energy is absorbed b. Exothermic

6.2

Describing Chemical Reactions Guided Reading and Study Use Target Reading Skills a. A chemical equation is a short, easy way to show a chemical reaction. b. A substance you have at the beginning of a reaction is a reactant. c. A new substance produced in a reaction is a product. d. Conservation of mass means that during a chemical reaction, matter is not created or destroyed. e. In an open system, matter can enter from or escape to the surroundings. f. In a closed system, matter is not allowed to enter or leave. g. A coefficient is a number in a chemical equation telling you how many atoms or molecules of a

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Jerry A. Bell



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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text Prentice Hall Science Explorer Michael J. Padilla, 2002 **Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations

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Jerry A. Bell, 1967 This forward looking laboratory text includes thirty three experiments designed to serve high level introductory chemistry courses It is intended in part to give the students practice in techniques not only those generally considered the techniques of chemistry but also those of mathematics However the only prerequisite is a good background in high school mathematics Ideally suited for use with University Chemistry or College Chemistry by Bruce Mahan this book follows the physical chemistry approach of these and most other modern chemistry texts maintaining the same flavor emphasis and degree of sophistication All experiments are open ended and include suggestions for following them up There is considerable emphasis on the involvement of the student in planning his own procedures and quantitative thinking

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