

# Physical properties of solutions

Even though “like dissolves like”, the presence of **solute molecules in the solvent**

**disrupts the bulk solvent intermolecular forces** to some extent.

**This change in forces** means that the **physical properties of solutions**,

like freezing and boiling points, **are slightly different than those of the pure solvent**.

# Physical Properties Of Solutions Concept Review

**WJ Hussar**



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