

Vicinal dibromide formation:

2. In solvent like H.O.



Halohydrin formation

- + It is a regionelective reaction.
- The nucleophilic attack of the water occurs at the more substituted carbon atom.

2. With Alkymes

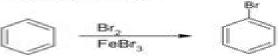
- · It is an anti-addition reaction.
- 1. In solvent like CCI, CH,CI, etc.



Tetrabormide formation

OH

3. With Aromatic Compounds



4. Photo halogenation of alkanes

5. Alpha Bromination of carbonyl compounds

6. Oxidation of Monosaccharides

- Bromine water is a mild oxidizing agent.
- selectively exidizes the aldehyde group (for example CHO group of glucose) to a carboxylic acid.

HO

OH

OH

7. Haloform Reaction

 converts methyl ketones to Haloform (CHX₃(X=Cl,Br,I)) and salt of carboxylic acids.

8. The Hoffmann Bromamide degradation

 It converts primary amide to a primary amine with one less carbon atom.

9. Hell-Volhard-Zelinski (or HVZ) reaction

5. CN

Cyanide Ion and Nitrile Group

1. Conversion of Nitriles to a Primary Amine

2. Conversion of Nitriles to Aldehydes

3. Reaction of Nitriles with Grignard reagent

4. Hydrolysis of Nitriles to Carboxylic Acids

Organic Chemistry Reagent Guide

Ochem Reagent Guide

Julia Schneider

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