



SOUTH AFRICAN MILITARY HEALTH SERVICES

BACKGROUND

The Department of Defence is proud to offer this opportunity to the South African youth with the main purposes of promoting and enhancing the concept of Nation building. We are also serving National interest by encouraging our youth to develop a common South African identity through exposure to military norms and customs. The emphasis will mainly be on discipline development, empowerment and global competitiveness. The SA Military Health Services is offering young South African citizens an excellent opportunity to serve in uniform over a 2 year period. These training and development opportunities will be constituted by means of a voluntary Military Skills Development System (MSDS) period of two years in order to equip young South Africans with market related skills.

GENERAL TRAINING

Potential Applicants may apply for the following two year training opportunities.

- **Animal Health General Training (Dog handlers and Veterinary Orderlies)** – Animal Health perform basic animal care, their primary duties include assisting with surgical procedures, dressing wounds and collecting specimens. The SAMHS is also responsible for the training of the SANDF Dog Handlers and recruits can be trained in this field. Graduates, Learners that completed Grade 12 or learners still busy with Grade 12 may apply.
- **Chemical Biological Radiation Defence (CBRD)** – This mustering involves activities and actions to defend against Chemical Biological Radiation weapons. This includes protection, detection and decontamination.
- **Physical Training Instructors (PTGI)** – This mustering is responsible for maintaining the fitness levels required for all uniform members. PTGI training takes place at the Joint Physical Training Sport and Recreation School in Pretoria. The duration of this learning opportunity is approximately 3 months. Applicants must be proficient in English.
- **Patient Administration** – This mustering is responsible for the administration of patients including reception personnel, account administration, maintenance of patient files and capturing of patient records electronically. Computer literacy, good communication and interpersonal skills are essential.

STUDY OPPORTUNITIES

Subsequent to the reporting of successful Applicants for military training, recruits can then be considered for the under mentioned developmental opportunities and tertiary training and will be selected for the said opportunities during the two year Military Skills Development System (MSDS) training. **PLEASE NOTE** that training in any of these fields cannot be guaranteed and will be subject to availability and further selection criteria.

- **PHARMACY ASSISTANT** – The pharmacy assistant training program is designed to provide broad knowledge of pharmacy practice with the aim of assisting in the performance of daily repetitive tasks and provide support to the pharmacist in the practice of a pharmacy. Applicants who wish to join this mustering need Grade 12 certificate or learners still busy with Grade 12 with English Level 4, Mathematics Level 4, Life Science Level 4 and Physical Science Level 4 as compulsory subjects.
- **MBCHB** – The degree qualifies successful students to be medical doctors. Subsequent to successful completion of the degree, Medical Doctors are employed mainly in Military Hospitals, Sickbays and Clinics that may include facilities in rural areas. Applicants must also note that the SAMHS deploy medical doctors for support to military operational deployments including peacekeeping in Africa. Applicants will need a Grade 12 certificate or learners still busy with Grade 12 with university exemption, English Level 6, Mathematics Level 6, Physical Science Level 6 and Life Science Level 6 as compulsory subjects, and a minimum APS of 38.
- **4 YEAR NURSING DIPLOMA** – Nursing is a healthcare profession focused on the care of individuals, families and communities so they may attain, maintain or recover optimal health and quality of life from conception to death. Applicants will need a Grade 12 certificate or learners still busy with Grade 12 with university exemption, English Level 4, Mathematics Level 4, and Life Science Level 4 as compulsory subjects, and a minimum APS of 24.
- **VETERINARY SCIENCE (VETERINARY SURGEON)** – The degree qualifies successful students to be Veterinary Surgeons. Applicants will need a Grade 12 certificate or learners still busy with Grade 12 with university exemption, English Level 6, Mathematics Level 6, Physical Science Level 6 and Life Science Level 6 as compulsory subjects, and a minimum APS of 36.

ENQUIRIES AND APPLICATIONS

Administrative enquiries can be directed to the SAMHS Recruitment Office at (012) 671 517/5428. If you are interested in the Military Skills Development System in the SA Military Health Services, please complete the coupon with the requested documentation enclosed and post it to: SAMHS Recruitment Office, Department of Defence, SA Military Health Services, Private Bag X100, Centurion, 0045.

APPLICANTS WILL BE SUBJECTED TO PSYCHOMETRIC TESTING, MEDICAL EVALUATION, VETTING ON CRIMINAL RECORD AND VERIFICATION OF QUALIFICATIONS DURING THE SELECTION PROCESS.

Please note: All applicants must comply with the following:

- Applicants must comply with health assessment requirements for appointment in the SANDF.
- Applicants must be accepted by a Military Selection Board and ascribe to the Military Code of Conduct.
- Applicants must be younger than 23 years of age and Graduates up to the age of 25.
- Applicants must be SA Citizens (compulsory).
- Applicants must have no record of criminal offence or any pending case.
- Applicants must be willing to undergo Basic Military Training for a period of 6 months, further Functional Training and serve in uniform within the RSA and abroad.
- Applicants must be willing not to be area bounded.

THIS APPLICATION FORM MUST BE SUBMITTED WITH THE FOLLOWING CERTIFIED DOCUMENTS:

1. Copy of ID
2. Copy of Gr 12 Certificate
3. A Short CV
4. Copy of highest achieved qualification (eg Degree or National diploma)

Samhs Bursaries 2016

SA Dillow



Samhs Bursaries 2016:

This book delves into Samhs Bursaries 2016. Samhs Bursaries 2016 is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Samhs Bursaries 2016, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Samhs Bursaries 2016
- Chapter 2: Essential Elements of Samhs Bursaries 2016
- Chapter 3: Samhs Bursaries 2016 in Everyday Life
- Chapter 4: Samhs Bursaries 2016 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Samhs Bursaries 2016. This chapter will explore what Samhs Bursaries 2016 is, why Samhs Bursaries 2016 is vital, and how to effectively learn about Samhs Bursaries 2016.
3. In chapter 2, the author will delve into the foundational concepts of Samhs Bursaries 2016. This chapter will elucidate the essential principles that must be understood to grasp Samhs Bursaries 2016 in its entirety.
4. In chapter 3, this book will examine the practical applications of Samhs Bursaries 2016 in daily life. This chapter will showcase real-world examples of how Samhs Bursaries 2016 can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Samhs Bursaries 2016 in specific contexts. This chapter will explore how Samhs Bursaries 2016 is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Samhs Bursaries 2016. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Samhs Bursaries 2016.

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq} + \text{CrO}_4^{2-} \text{aq} \rightarrow \text{Ag}_2\text{CrO}_4 \text{s}$

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web chemical equation in which all dissolved ionic reactants and products including spectator ions are explicitly represented by formulas for their dissociated ions molecular equation chemical equation in which all reactants and products are

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web jan 29 2020 the ionic form of this reaction is $2\text{Na}^+ \text{aq} + 2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq} \rightarrow 2\text{Na}^+ \text{aq} + \text{SO}_4^{2-} \text{aq} + \text{CuCl}_2 \text{s}$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web mar 11 2014 solution ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ the spectator ions are Cl^- and Na^+ net ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ $\text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ here is a video on spectator ions *chapter 8 04 ionic equations chemistry libretexts* - Mar 06 2023

web the six $\text{NO}_3^- \text{aq}$ ions and the six $\text{Na}^+ \text{aq}$ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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 $\text{AgNO}_3(\text{aq}) + \text{KCl}(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{KNO}_3(\text{aq})$ total ionic $\text{Ag}^+(\text{aq}) + \text{NO}_3^-(\text{aq}) + \text{K}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{K}^+(\text{aq}) + \text{NO}_3^-(\text{aq})$ net ionic $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s})$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l})$ the spectator ions in $\text{KOH}(\text{aq})$ and $\text{HNO}_3(\text{aq})$ are $\text{K}^+(\text{aq})$ and $\text{NO}_3^-(\text{aq})$ how to identify and list the

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web the $\text{Ag}^+(\text{aq})$ and $\text{Cl}^-(\text{aq})$ ions become $\text{AgCl}(\text{s})$ but the $\text{Na}^+(\text{aq})$ ions and the $\text{NO}_3^-(\text{aq})$ ions stay as $\text{Na}^+(\text{aq})$ ions and $\text{NO}_3^-(\text{aq})$ ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both

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