

Question	Answer	Marks	Guidance
5 (a)	<p>F_2 forces mark F_2 has van der Waals' (forces) OR F_2 has induced dipole attractions OR interactions OR F_2 has temporary OR instantaneous dipole(-dipole) attraction OR interactions ✓</p> <p>HCl forces mark HCl has permanent dipole(-dipole) attractions OR interactions ✓</p> <p>Comparison of strength of forces between molecules mark intermolecular force in HCl is stronger than that in F_2 OR permanent dipoles are stronger (than induced dipoles) ✓</p> <p>Boiling point mark more energy is required to break stronger (intermolecular) forces ✓</p>	4	<p>Use annotations ie ticks crosses ECF ^ etc for this part</p> <p>ALLOW vdWs for van der Waals' IGNORE F_2 has covalent bond for this mark IGNORE F_2 has 'intermolecular forces'</p> <p><i>Quality of written communication: 'dipole(s)' spelled correctly and used in context for the second marking point</i> IGNORE HCl has 'intermolecular forces' IGNORE van der Waals' forces in HCl/ DO NOT ALLOW hydrogen bonding DO NOT ALLOW ionic bonding</p> <p>Look for strength of force comparison anywhere in the answer ALLOW ECF for hydrogen bonding in HCl being stronger than the stated intermolecular forces in F_2 BUT DO NOT ALLOW this mark if HCl or F_2 has covalent bonds broken OR if HCl has ionic bonds broken (the question asks for forces between molecules) IGNORE HCl has stronger van der Waals' (forces) than F_2 (as they both have the same number of electrons)</p> <p>DO NOT ALLOW fourth mark if covalent bonds are broken in HCl or F_2 OR if ionic bonds are broken in HCl</p> <p>IGNORE 'heat' but ALLOW 'heat energy'</p>

Ocr F321 Jan 2013 Paper

Mike Smith



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Cambridge International AS and A Level Chemistry Peter Cann, Peter Hughes, 2015-03-06 Endorsed by Cambridge Assessment International Education for full syllabus coverage Foster a deeper understanding of theoretical concepts through clear guidance and opportunities for self assessment throughout covers the entire Cambridge International AS A Level Chemistry syllabus 9701 Navigate the different routes through the course with ease with clearly divided sections for AS and A Level Focus learning with learning outcomes clearly defined at the beginning of each section Test knowledge and understanding with past paper and exam style questions Address the Key Concepts in the syllabus which are clearly highlighted throughout the course The Revision and Practice CD included with every Student's Book provides interactive tests summaries of each topic and advice on examination techniques

OCR A Chemistry AS Student Unit Guide: F321 New Edition Atoms, Bonds and Groups Mike Smith, 2012-08-31 Written by a former senior examiner Mike Smith this OCR A AS Chemistry Student Unit Guide is the essential study companion for Unit F321 Atoms Bonds and Groups This full colour book includes all you need to know to prepare for your unit exam clear guidance on the content of the unit with topic summaries knowledge check questions and a quick reference index examiner's advice throughout so you will know what to expect in the exam and will be able to demonstrate the skills required exam style questions with graded student responses so you can see clearly what is required to get a better grade

OCR Revise AS Chemistry Mike Wooster, Helen Eccles, 2008 Written by experienced examiners this title offers tips on tricky areas and questions on each topic

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- Chapter 3: Ocr F321 Jan 2013 Paper in Everyday Life
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