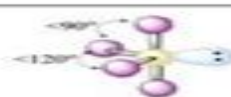


Molecular Geometry Chart

# of Electron Groups	Number of Lone Pairs	Electron Pair Arrangement	Molecular Geometry	Approximate Bond Angles
2	0	linear		180°
3	0	trigonal planar		120°
	1	bent		$<120^\circ$
4	0	tetrahedral		109.5°
	1	trigonal pyramid		$<109.5^\circ$ ($\sim 107^\circ$)
	2	bent		$<109.5^\circ$ ($\sim 105^\circ$)
5	0	trigonal bipyramidal		$90^\circ, 120^\circ$
	1	see-saw		$<90^\circ, <120^\circ$
	2	T-structure		$<90^\circ$
	3	linear		180°
6	0	octahedral		$90^\circ, 90^\circ$
	1	square pyramidal		$90^\circ, <90^\circ$
	2	square planar		90°

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