

Polyatomic Ions

Can a group of atoms have a charge?

Why?

Do you know you eat a lot of "-ates"? Next time you look at a food label, read the ingredients and you will likely find a number of ingredients that end with "-ate," such as sodium phosphate or calcium carbonate. Did you ever wonder what the chemical formulas of these ingredients look like? In this activity we will explore polyatomic ions, which are groups of atoms that carry a charge. These ions are found in our food ingredients, natural waterways, and many other chemical compounds you encounter every day.

Model 1 – Types of Ions

Monatomic Ions	Nitride 	Sulfide 	Chloride 
Polyatomic Ions	Nitrate 	Sulfate 	Ammonium 
	Nitrite 	Sulfite 	Hydroxide 

1. Use Model 1 to complete the table below.

Name of Ion	Nitride	Nitrate	Sulfate	Sulfite	Ammonium
Charge on Ion	-3	-1	-2	-2	+1
Type and Number of Atoms	1 nitrogen	1 nitrogen 3 oxygen	1 sulfur 4 oxygen	1 sulfur 3 oxygen	1 nitrogen 4 hydrogen
Chemical Formula	N^{3-}	NO_3^{1-}	SO_4^{2-}	SO_3^{2-}	NH_4^{1+}

Polyatomic Ions Pogil Activities For High School Chemistry Answers

Peter Scott, Rachael Enriquez Porter



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Dr. Salm's Chemistry Problem Drill Book Robert J. Salm, Robert J. Salm Robert J. Salm Ph.D., 2005 This drill book contains many common problem types that are asked in General Chemistry classes in High School and College This work will give you practice with the major problem types as you prepare for finals and standardized tests **Polyatomic Ions for Surface**

Analysis and Modification Erick Ryan Fuoco, 2003 **The Particulate Nature of Polyatomic Ions**, 2007 Quick Practice Reproducibles for Chemistry I Tim McDowell, 2012-05-18 A collection of reproducible practice for high school chemistry and middle school physical science Topics include Metric Units Chemical Symbols for elements Atomic Numbers Isotopes Chemical Formulas and Polyatomic Ions **Study Guide for Introductory Chemistry** Iris Stovall, 1990

Experimental Studies of Polyatomic Ion Interactions with Clean and Adsorbate Covered Metal Surfaces Samuel B. Wainhaus, 1997 **Study Guide for Hein and Arena's Foundations of College Chemistry** Peter Scott, Rachael Enriquez Porter, 1999-04-18

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