

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) Titration with NaOH

a. CH_3COOH volume 20.00 mL and NaOH 0.1000 M

b. CH_3COOH volume 20.00 mL and NaOH 0.1000 M

c. CH_3COOH and NaOH prepared for titration buffer
 0.1000 M and 0.1000 M

Record Values of pH and pH for each and pH for each

pH 4.75 pH 4.75 pH 4.75
 pH 4.75 pH 4.75 pH 4.75

Species	Add NaOH , Base (pH or NaOH pH)	Acids with NaOH (pH or NaOH)	Conductivity (pH or NaOH)
CH_3COOH	pH	pH	pH
CH_3COO^-	pH	pH	pH
NaCl	pH	pH	pH
NaOH	pH	pH	pH

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