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**M Carnoy**



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web temperature and pressure hints density g l use the ideal gas law to solve for moles first a 0 35 g l b 2 14 g l c 2 86 g l first use the ideal gas law and recall that density is g l 101 3 x 1 l n x 8 31 x 273 therefore n 0 045 moles density is g l so convert moles to grams by multiplying by the molar mass of so 2

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web the observed behavior of gases embodied in the empirical gas laws leads to a series of equations that can be summarized by a single equation of state called the ideal gas law equation this shows the relationship between a gas's pressure  $p$  temperature  $T$  volume  $V$  and amount in moles  $n$

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web each gas in a mixture of gases exerts its own pressure independently of the other gases present therefore the pressure of each gas within a mixture is called the partial pressure of the gas dalton's law of partial pressures states that the total pressure of a mixture of gases is equal to the sum of the partial pressures of the component gases

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