d. Secondary passive transport protein

10. An intermediate compound X is formed during the production of urea through ammonia/carbon dioxide technology. What is the molecular formula of X?

- a. NH₂COONH₂
- b. NH₂COONH₄
- c. NH₄COONH₂
- d. NH₄COONH₄
- 11. Which of the following are recycled in the manufacture of sodium Carbonate by Solvay's process?
 - CO₂ and NH₄Cl
 - b. CO₂ and NH₃
 - c. NaCl and CaO
 - d. NaCl and NH₃

Group 'B'

Give short answer to the following questions.

 $(8 \times 5 = 40)$

- An element X has 2 electrons in K shell, 8 electrons in L shell and 5 electrons in M shell.
 - Identify the element X and write the number of protons and electrons in it. [3]
 - Size of X- - ion is greater than that of X atom though both contain the same number of protons. Give reason [1]
 - iii. c. Write down the formula of one of the compounds of X where X is in -3 oxidation state. [1]

\mathbf{OR}

Know -how about ionization energy (IE) of elements is crucial aspect in the study of chemical bonding whether they form ionic or covalent bond. The first ionization energies of period second elements are given below

of elements				
onization energy(kJ/mol)				

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William Shurtleff, Akiko Aoyagi

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