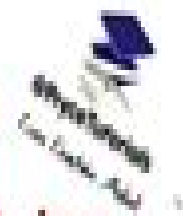


Volumetric Analysis Calculations



30.0 cm³ of 0.100 mol dm⁻³ NaOH reacted completely with 25.0 cm³ of H₂SO₄ in a titration.

Calculate the concentration of H₂SO₄ in mol dm⁻³?

Step 2: Find no' of moles in NaOH



✓ $c = 0.1 \text{ mol dm}^{-3}$

$V = 25 \text{ cm}^3$

✓ $V = \underline{30 \text{ cm}^3}$



Using $n = c \times v$

$$\begin{aligned} n(\text{NaOH}) &= 0.1 \times \frac{30}{1000} \\ &= 0.003 \text{ mol} \end{aligned}$$

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